# Accepted Manuscript

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PII: S0360-5442(19)30265-8

DOI: https://doi.org/10.1016/j.energy.2019.02.070

Reference: EGY 14712

To appear in: Energy

Received Date: 12 December 2018

Revised Date: 17 January 2019

Accepted Date: 10 February 2019

Please cite this article as: Ying J, Eimer DA, Mathisen A, Brakstad F, Haugen HA, Ultrasound intensify CO<sub>2</sub> desorption from pressurized loaded monoethanolamine solutions. II. Optimization and cost estimation, *Energy* (2019), doi: https://doi.org/10.1016/j.energy.2019.02.070.

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# Ultrasound Intensify CO<sub>2</sub> Desorption from Pressurized Loaded Monoethanolamine Solutions

II. Optimization and Cost Estimation

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# ABSTRACT

Optimization and cost estimation were performed for the use of ultrasound for intensifying  $CO_2$  reboiler stripping from lean monoethanolamine (MEA) aqueous solutions at 1.0 barg. This work was based on typical industrial reboiler operation conditions. Experiments were run by intermittently applying ultrasound for  $CO_2$  stripping. A multi-variable data analysis was employed to explain the results and find the optimum for ultrasound operation. The results show that the  $CO_2$  stripping rate by ultrasound is more than 3 times than heat only in the reboiler. A normalized specific energy consumption was defined based on the classic industrial case (4.2 kJ/kg  $CO_2$ ), and the normalized specific energy consumption 3.6 MJ/kg  $CO_2$  was deduced, showing 14% energy saving. Cost estimations have been conducted using Aspen Plus V9.0 and Aspen In-plant Cost Estimator for the industrial cases with/without the assistance of ultrasound. Total capture cost is 60.2 EUR/t  $CO_2$  and cost saving is 19% when the  $CO_2$  loading of the lean MEA solution can be further decreased to 0.20 mol/mol from 0.44 mol/mol assisted by ultrasound.

# 1. INTRODUCTION

In current amine-based  $CO_2$  capture plants the operational cost is the major cost element representing approximately 2/3rds of the specific  $CO_2$  capture cost, with the largest single contributor being the energy consumption in the desorber reboiler. There is a significant potential to reduce the total cost of  $CO_2$  capture by improving the  $CO_2$  stripping technology for this type of plant.

Ultrasound enhanced  $CO_2$  stripping could unlock  $CO_2$  capture cost savings. As a mature technology, ultrasound is widely used in other fields [1-3]. Recently, ultrasound was introduced in gas purification for stripping acid gas from loaded solutions, for absorbent regeneration, and reducing degradation of absorbents [4-7].

The major mechanisms for improving the mass transfer are the millions of cavitation bubbles created by ultrasound due to cavitation and nucleation effects. This results in increased interfacial area between gas and liquid. Bubbles form more easily and the activation energy for surface diffusion decreases [8]. The application of ultrasound intensifies the conversion of free  $CO_2$  in the liquid phase to  $CO_2$  (g). Thus, the chemical reaction freeing  $CO_2$  is accelerated as the freed  $CO_2$  is more efficiently removed from the liquid solution. The force driving  $CO_2$  to the bubbles is great because the cavitation bubbles are in a state of vacuum at the beginning. Further effects of ultrasound that enhance mass transfer include formation of micro-streams and vortices, enhancement of bubble growth, acoustic streaming and rectified diffusion [9, 10]. Coalescence of small bubbles will further enhance the gas' ability to rise to the liquid surface, which can also be accelerated by ultrasound through Bjerknes forces [11].



Figure 1. Four potential positions (highlighted in red) for installation of ultrasound to strip CO<sub>2</sub> in an amine-based CO<sub>2</sub> capture plant.

In a typical MEA-based  $CO_2$  capture plant, as illustrated in Figure 1, there are four potential locations where ultrasound could be applied to enhance stripping of  $CO_2$ , these are indicated in the flowsheet by four ellipses. Placement in the reboiler (US-4 in Figure 1) is believed to be the most optimal for ultrasound. The reason being that it is in the reboilers that the absorbent is heated by

steam to drive the reactions releasing the chemically bound  $CO_2$ . From our previous work [12-15] it was found that most ultrasound energy were used for heating rather than forming bubbles to drive free  $CO_2$  (the  $CO_2$  in liquid phase) off when the desorption temperature was low, and this resulted in using electrical energy to replace heat. The release of gas is one of the rate-limiting factors in the desorption process. The reboiler in particular, although there is significant bubbling already, contains liquid with a "rest content" of free  $CO_2$  that struggles to reach the gas phase for reasons stated above. If ultrasound is applied efficiently, limited to enhance the release of  $CO_2$  (aq.) in the absorbent into  $CO_2$  (g) and not heat the solution, the total energy consumption in the reboiler should be reduced. Therefore, to save energy, the best position of ultrasound application in an industrial process could be the position US-4 in Figure 1. In this place, ultrasound could enhance  $CO_2$  stripping from the lean MEA solutions, and achieve leaner MEA solutions than usual. For example, the loading could be reduced from 0.25 mol/mol down to 0.20 mol/mol assisted by ultrasound. This will result in an increased cycling capacity of the absorbent. Hence, the same  $CO_2$  capture rate could be achieved with a lower circulation rate, resulting in a lower energy consumption, because the consumption of steam, energy for pumps and area of heat exchangers is reduced.

In the current work, in order to provide useful information for the industrial applications, the operating conditions of ultrasound intensification of  $CO_2$  stripping was optimized for this lab-scale reboiler based on the results of previous parameter investigation [15], and a cost estimation was performed and compared to an industrial base case. To achieve this goal, an intermittent ultrasound operation was adopted in the experiments at varying pressures up to 1.5 barg at the boiling point of a lean MEA aqueous solution (0.25 mol/mol). Aspen Plus V9.0 and Aspen In-plant Cost Estimator were employed for the cost estimations, and an MEA-based  $CO_2$  capture from a Combined Cycle Gas Turbine (CCGT) natural gas based power plant was chosen as the industrial base case for comparison.

# 2. EXPERIMENTAL SECTION

#### 2.1. Reagents and Solutions Preparation

Reagent grade MEA with mass fraction purity  $\geq 99.5\%$  from Merck and was used without further purification. Deionized water (purified with an ELGA Purelab Prima 7, resistivity more than 0.05 MΩ-cm) and MEA were mixed in a tank to produce 0.02 m<sup>3</sup> of 30 wt% MEA aqueous solution. This solution was loaded by bubbling CO<sub>2</sub> (purity  $\geq 99.995\%$ , produced by AGA Gas, Norway) through 3 sinters in the bottom of the tank with a CO<sub>2</sub> flow rate of  $3.3 \times 10^{-5}$  m<sup>3</sup>/s to prepare the desired loading. The CO<sub>2</sub> loading was determined by a density method [14] before the experiments. If the loading was not as expected, e.g. higher than 0.25 mol/mol, more unloaded fresh 30 wt% MEA solution was added and mixed in the solution to obtain the desired loading.

#### 2.2. Experimental Equipment and Procedures

The test rig is same as the rig used in the previous parameter investigation work [15]. To avoid pressure surges in the reboiler when ultrasound was running, the pressure was controlled by a needle valve (8) instead of the on/off solenoid valve used in the previous work. During the experiments, once the pressure in the cell reached the desired pressure, the needle valve (8) was opened and adjusted manually to a stable gas flow. The modified flowsheet of the rig is given in Figure 2. More details of the operational procedure are given in our previous work [15].



Figure 2. Schematic diagram of the rig used to study ultrasound-enhanced CO<sub>2</sub> stripping from amine solutions.
1, Rich amine feed tank; 2, Liquid flow meter; 3, Preheat unit; 4, Glass kettle reboiler; 5, Needle valve; 6, Lean amine receiving tank; 7, Gas-liquid separator; 8, Regulatable needle valve; 9, CO<sub>2</sub> flow meter; 10, Various sensors (P, T) in liquid and gas; 11, Ultrasound unit; 12, Electric steam generator; 13, Steam trap; 14, Condensed steam receiving tank; 15, balance; 16, High speed camera; 17, Pump. (Blue lines are liquid phase and Green lines are gas phase)

# 2.3. Experimental Matrix Design

The influence of various parameters such as pressure, temperature, flow rate of liquid, CO<sub>2</sub> loading, frequency and intensity of ultrasound have been previously investigated and modelled [15]. The results suggested that higher CO<sub>2</sub> loading and frequency of ultrasound are strongly positive with regard to energy saving, and temperature (or pressure), liquid flow rate, intensity of ultrasound are weakly negative with respect to energy saving. Based on our previous investigation, a 28 kHz multi-surface sonotrode and 100 % intensity output of ultrasound were chosen and fixed in this work. The liquid flow rate was not a significant variable and was set to  $1.7 \times 10^{-5}$  m<sup>3</sup>/s. CO<sub>2</sub> loading and pressure are very important variables. However, to keep the same conditions as a reboiler of a typical industrial MEA-based CO<sub>2</sub> capture process, CO<sub>2</sub> loading was fixed to 0.25 mol/mol, pressure was set to 1 barg and the steam temperature was set to 130 °C in this work. The ultrasound running time has a significant effect on the CO<sub>2</sub> stripping rate and energy consumption, and it represents the variable to be optimized.

In this work, ultrasound application was intermittent with an on-time ( $t_{on}$ ) and an off-time ( $t_{off}$ ). These were varied for the purpose of energy optimization. The experimental matrix was constructed using a Central Composite Design [16] generated by Design Expert V.9.0.6.2 from Stat-Ease. The

design matrix is shown in Table 1. For each variable, a low, a middle and a high level were decided on, and the experiments were carried out such that all variable combinations were systematically covered. Three duplicate experiments on average values of the variables (i.e. No. 3, 7 and 11) were used to quantify the random variation. In the design, the parameter, on-time fraction of ultrasound in a period ( $\varphi$ ), was proposed to analyze the efficiency of ultrasound on CO<sub>2</sub> desorption.

Pup NO	Factor 1	Factor 2	Ref. factor 1	Ref. factor 2
Kull NO. –	$t_{\rm on}$ (s)	$t_{\rm off}(s)$	$\varphi = t_{on}$ / period	Cycle period (s)
1	3	1	75%	4
2	5	5	50%	10
3	3	3	50%	6
4	5	1	83%	6
5	3	5	38%	8
6	1	1	50%	2
7	3	3	50%	6
8	1	3	25%	4
9	5	3	63%	8
10	1	5	17%	6
11	3	3	50%	6

Tabla 1	The	designed	test matrix	for	ontimization	<b>`</b>
rable r.	rne	designed	test matrix	IOL	opunnization	1

The predicted optimum conditions can be obtained by both numerical and graphical evaluations, and verification experiments at the suggested optimum were made by three replicates to make a better verification.

# 3. RESULTS AND DISCUSSION

# 3.1. Definition of respondents / parameters

To explain the results clearly, three dependent variables related to specific energy consumption and one dependent variable with respect to  $CO_2$  stripping kinetics are defined.

#### 3.1.1. Specific energy consumption in reboiler

The first response used in this analysis, specific energy consumption,  $E_s$ , including two cases:  $E_{s,H}$  for heat only and  $E_{s,US}$  for heat + ultrasound assistance respectively. Both, in unit MJ/kg CO<sub>2</sub>, are defined as

$$E_{\rm s,H} = \frac{H_{\rm st} \times \gamma - H_{\rm cw}}{A_{\rm CO2}} \quad \text{and} \quad E_{\rm s,US} = \frac{H_{\rm st} \times \gamma + H_{\rm US} - H_{\rm cw}}{A_{\rm CO2}} \tag{1}$$

where  $A_{CO2}$  is the total rate of CO<sub>2</sub> stripped (in unit mg/s),  $H_{st}$  is the energy input into the reboiler from steam,  $H_{US}$  is the ultrasound energy input,  $\gamma$  is the energy efficiency of steam heat input. A blank experiment (water in the reboiler) at room pressure when the steam temperature was set to 130 °C, gave 88% energy efficiency.

#### 3.1.2. Normalized Specific energy consumption

In this work, the vapor (mainly water) of the MEA solution from the reboiler was condensed and not taken to a desorption column for further CO<sub>2</sub> recovery from a richer solution as in an industrial setting. For this reason, the specific steam consumption (with respect to CO<sub>2</sub>) observed in the experimental rig cannot be compared directly to the standard industrial yardstick of 4.2 MJ/kg CO<sub>2</sub>, (which varies from 3.2 to 5.5 MJ/kg CO<sub>2</sub> reported by literature [17, 18]). To enable comparison to an industry case, a normalization calculation must be used. The hypothesis is as follows;

- 1. The vapor produced in the lab-scale reboiler could in principle be used for further  $CO_2$  stripping in a desorption column (where most of the  $CO_2$  stripping is normally accomplished). By definition the stripping rate of  $CO_2$  by the vapor in the assumed desorption column is  $A_{CO2,v,H}$  or  $A_{CO2,v,US}$ .
- 2. The energy consumption from pumps etc. is neglected. (This means that the proposed estimation is conservative because the power of the pumps and size of the exchangers can be reduced due to a leaner solution being produced by introducing ultrasound.)
- 3. The specific energy consumption in lab scale can be scaled up to industrial case linearly.

Based on these assumptions, the following calculation can be made. When the solution is treated by heat only, the rate of CO<sub>2</sub> stripping is  $A_{CO2,v,H}$ , and this is defined by the equation:

4.2 (MJ/kg CO<sub>2</sub>) = 
$$\frac{H_{st} \times \gamma}{A_{CO2} + A_{CO2,v,H}}$$
 (2)

where the value 4.2 MJ/kg CO<sub>2</sub> is the typical specific energy consumption in a CO<sub>2</sub> capture plant based on 30 wt% aqueous MEA solution,  $H_{st}$  is the energy input by steam heat,  $A_{CO2,v,H}$  is the assumed CO<sub>2</sub> stripping rate by the vapor (from the reboiler) into the assumed desorption column with heat treatment only,  $A_{CO2}$  is the CO<sub>2</sub> stripping rate observed in the reboiler rig,  $\gamma$  is the energy efficiency of steam heat input.  $A_{CO2,v,H}$  can be calculated by equation (2) when  $H_{st}$  and  $A_{CO2}$  are measured.

The CO<sub>2</sub> stripping rate in the assumed desorption column by vapor when ultrasound is introduced is defined as  $A_{CO2,v,US}$ . In this case, vapor is produced by heat and ultrasound in the reboiler when ultrasound is applied. Assuming that the CO<sub>2</sub> stripping rate in the column is in proportion to the vapor flux, then  $A_{CO2,v,US}$  can be defined as follows (in unit mg/s),

$$A_{\rm CO2,v,US} = A_{\rm CO2,v,H} \times \frac{W_{v,US}}{W_{v,H}}$$
 (3)

where  $W_{v,H}$  is the weight of vapor from the reboiler produced by heat in unit time, and  $W_{v,US}$  is the weight of vapor from the reboiler produced in unit time when steam heat and ultrasound are applied simultaneously.

The normalized specific energy consumptions,  $\bar{E}_{s}$ , including  $\bar{E}_{s,H}$  or  $\bar{E}_{s,US}$ , can be calculated by equations (4) and (5).

For heat only,

$$\overline{E}_{s,H} = \frac{\text{Total energy input}}{\text{Total stripped CO}_2} = \frac{H_{st} \times \gamma}{A_{CO2} + A_{CO2,v,H}} , \quad (\text{MJ/kg CO}_2)$$

For ultrasound application (heat + ultrasound),

$$\overline{E}_{s,US} = \frac{\text{Total energy input}}{\text{Total stripped CO}_2} = \frac{H_{st} \times \gamma + H_{US}}{A_{CO2} + A_{CO2,v,US}} , \quad (\text{MJ/kg CO}_2)$$
(5)

#### 3.1.3. Energy saving

The energy saving  $(\eta)$  due to using ultrasound in the process is defined as

$$\eta = \frac{E_{\rm s,H} - E_{\rm s,US}}{E_{\rm s,H}} \times 100\%$$
(6)

The above defined energy saving is based on the reboiler as the control element. If based on the control element including a reboiler and a desorption column, the normalized energy saving is

$$\overline{\eta} = \frac{\overline{E}_{s,H} - \overline{E}_{s,US}}{\overline{E}_{s,H}} \times 100\%$$
(7)

Where  $\bar{E}_{s,H}$  is the typical specific energy consumption of an MEA plant, i.e. 4.2 MJ/kg CO<sub>2</sub>.

# 3.1.4. Improvement of CO<sub>2</sub> stripping rate

The improvement of CO<sub>2</sub> stripping rate by ultrasound is defined as

$$\lambda = \frac{A_{\rm CO2} - A_{\rm CO2,H}}{A_{\rm CO2,H}} \times 100\%$$
(8)

Where  $A_{\text{CO2,H}}$  is the CO<sub>2</sub> stripping rate by heat only (in unit mg/s).

## **3.2.** Experimental Results

The results of the measurements are shown in Table 2, and illustrated in Figure 3 to Figure 6. The raw data of the measurements such as  $H_{st}$ ,  $H_{US}$ ,  $W_{v,H}$  and  $W_{v,US}$  are also listed in the table.

		Ul	trasound		Stea	m	Condensed vapor	C	O <sub>2</sub> stripping r	ate	Spe	ecific Ener	rgy Consumption	
No.	ton	$t_{\rm off}$	φ	$H_{\rm US}$	Mass	$H_{\rm st}$	$W_{\rm v,US}$ (or $W_{\rm v,H}$ )	$A_{\rm CO2}$	$A_{\rm CO2,v,H}$ (or $A_{\rm CO2,v,US}$ )	λ	$E_{s,US}$ (or $E_{s,H}$ )	η	$ar{E}_{ m s,US}$ (or $ar{E}_{ m s,H}$ )	$\overline{\eta}$
	s	s	%	J/s	g	J/s	mg/s	mg/s	mg/s	%	MJ/kg	%	MJ/kg	%
Heat	0	-	0%	0.0	22	249.8	58.8	9.5	50.0	(- )	12.8	-	4.2	-
1	3	1	75.0%	345.0	23.5	266.8	84.7	50.5	72.0	432%	8.5	34%	5	-19%
2	5	5	50.0%	230.0	23.3	263.0	84.2	54.0	71,5	468%	5.7	55%	3.9	7%
3	3	3	50.0%	230.0	23.2	262.0	83.5	37.0	71.0	289%	8.4	34%	4.6	-10%
4	5	1	83.3%	383.3	23.4	267.3	84.2	49.2	71.5	418%	9.5	26%	5.4	-29%
5	3	5	37.5%	173.3	22.4	254.3	79.3	24.8	67.5	161%	10.2	20%	4.6	-10%
6	1	1	50.0%	230.0	23.5	266.8	84.7	49.8	72.0	424%	6.3	51%	4.1	2%
7	3	3	50.0%	230.0	23.1	265.5	83.0	39.7	70.5	318%	7.9	38%	4.5	-7%
8	1	3	25.0%	115.0	22.1	251.0	78.2	26.2	66.5	176%	7.4	42%	3.9	7%
9	5	3	62.5%	288.3	23.3	264.7	83.5	43.2	71.0	355%	8.5	34%	4.8	-14%
10	1	5	16.7%	76.7	20.9	237.3	71.8	15.3	61.0	61%	10.2	20%	4.1	2%
11	3	3	50.0%	230.0	23.2	266.7	83.0	40.0	70.5	321%	7.8	39%	4.5	-7%
Opti.	1	2	33.3%	153.3	23.3	264.7	84.2	43.2	71.5	355%	5.4	58%	3.6	14%

 Table 2. The results of the test matrix for optimization

Note: Optimized run (Opti.) done based on the optimization work runs no. 1 -11.

From Figure 3, it can be seen that the CO<sub>2</sub> stripping rate increases significantly when assisted by ultrasound. The improvement of CO<sub>2</sub> stripping rate by ultrasound treatment has been observed to reach 300% compared to heat treatment only. This is because ultrasound can produce millions of cavitation bubbles, which greatly increase the interface area of gas and liquid, and also the microstreams and vortices produced by ultrasound can intensify the gas molecules' mass transfer in the liquid. It is found that the CO<sub>2</sub> stripping rate increases as the on-time fraction of ultrasound ( $\varphi$ ) increases, indicating that a longer time of ultrasound treatment results in a higher CO<sub>2</sub> stripping rate. This is reasonable because longer action time of ultrasound would produce more cavitation bubbles and then desorb more CO<sub>2</sub> from the solution. However, the increasing tendency of the CO<sub>2</sub> stripping rate becomes weak and flattens out when  $\varphi > 50\%$ . This is because the accumulated free CO<sub>2</sub> in the liquid decreases after ultrasound treatment, and the CO<sub>2</sub> production rate from carbamate is not fast enough, i.e., most of the free CO<sub>2</sub> is stripped by ultrasound at the early stage, and there is little free CO<sub>2</sub> in the solution to strip for the later stage. Free CO<sub>2</sub> tends to accumulate in the ultrasound off-time.



**Figure 3.** The CO<sub>2</sub> stripping rate (A<sub>CO2</sub>) as a function of the on ontime fraction of ultrasound ( $\varphi$ ) (the red line is a 2<sup>nd</sup> order polynomial regression)



It is noted that the experimental data are not a good fit to the 2<sup>nd</sup> order polynomial curve, the determination coefficient (R<sup>2</sup>) of the regression is only 0.77. For example, the dispersion (from 40 to 54 mg/s, the standard deviation = 6.7 mg/s) is high when  $\varphi = 50\%$ . This is because the use of  $\varphi$ , which incorporates both  $t_{on}$  and  $t_{off}$ , as the independent variable in this plot.

The CO<sub>2</sub> hypothetically stripped in the assumed desorption column depends on the amount of vapor from the reboiler. In Figure 4, the flow rate of the vapor from the reboiler as a function of  $\varphi$  can be seen. Similar to the relationship of CO<sub>2</sub> stripping rate and  $\varphi$ , the vapor flow rate increases with an increase of  $t_{on}$ , the R<sup>2</sup> of the regression is 0.96, and dispersion of the data is lower. This is because the vapor produced depends on the numbers of growing cavitation bubbles, which are

directly affected by  $t_{on}$ . However, because the accumulated free CO<sub>2</sub> in the liquid decreases along with the ultrasound application, especially when  $\varphi > 50\%$ , lots of cavitation bubbles cannot grow but collapse, and then the increased trend of the condensed vapor produced rate becomes weak as  $t_{on}$  increases.



on-time fraction of ultrasound ( $\varphi$ ) (red line is a 2<sup>nd</sup> order polynomial regression)



Figure 5 shows the  $E_s$  as a function of  $\varphi$ . It can be seen that the  $E_s$  decreases significantly with the increase of  $\varphi$  at first, and after reaches minimum value (when  $\varphi$  is in the range 40% - 50%) before increasing again. This indicates that the optimum condition for maximum energy saving is when  $\varphi$  is in the range 40% - 50%. However, it is noted that the experimental data are not consistent when  $\varphi$  is 50%, the standard deviation is about 1.03 MJ/kg CO<sub>2</sub>, implying that interaction between  $t_{on}$  and  $t_{off}$  cannot be neglected (this will be discussed in the section of data analysis).  $E_s$  decreases as the on-time of ultrasound application is lower than 40 % mainly because:

- 1. Cavitation and nucleation effects of ultrasound leads to lower energy consumption to form bubbles compared to heating only;
- 2. Micro-streams and vortices produced by ultrasound enhance the gas molecules mass transfer in the liquid, and they cause a lower concentration of free  $CO_2$  in the liquid. This is beneficial for  $CO_2$  conversion reaction from carbamate.

The specific energy consumption starts increasing after  $\varphi \approx 40\%$ . This is because the accumulated free CO<sub>2</sub> in the liquid decreases after ultrasound was applied for a few seconds, and the CO<sub>2</sub> production rate from carbamate is not fast enough. Hence many cavitation bubbles collapse and disappear, this observation manifests that a part of the ultrasound energy input is wasted /converted into other forms such as heating of the liquid.

The normalized specific energy consumption  $(\bar{E}_s)$  as a function of  $\varphi$  is shown in Figure 6. It can be seen that  $\bar{E}_s$  decreases as  $\varphi$  increases of at beginning and reaches a minimum value when  $\varphi$  is in the range of 30% - 40% then starts increasing. The energy consumption increases fast after  $\varphi \approx$ 50%. Because the CO<sub>2</sub> stripped in a desorption column depends on the amount of vapor produced from the reboiler, the produced vapor becomes less after  $\varphi \approx$  50% (see Figure 4), and the free CO<sub>2</sub> in the solution becomes less at the later stage.

#### 3.3. Data Analysis and Optimization

Based on the results, a multivariable analysis for optimization was conducted. The respondent variable used in this optimization analysis is  $\bar{E}_s$ . The average  $\bar{E}_s$  for the replicates (exp. 3, 7 and 11) are 4.53 MJ/kg CO<sub>2</sub> (95% confidence interval), and the relative standard deviation (RSD) is 1.3%, implying that the experimental work has a good quality proof. For comparison, the effect on  $\bar{E}_s$  by varying  $t_{on}$  and  $t_{off}$  from 1 to 5 seconds, generates a variation in the  $\bar{E}_s$  in the interval (3.9 – 5.4). Statistically speaking, the F-value [16] of the models is 4.8, and the probability is below 4 % that such high F-values can be caused by noise only.

The mathematical model derived from the regression analysis is,

$$E_s = 3.332 + 0.517 \times t_{on} + 0.175 \times t_{off} - 0.106 \times (t_{on} \times t_{off})$$
(9)

 $R^2$  of the multivariable regression is 0.8, a comparison of predicted and measured values is shown in Figure 7.



Figure 7. Comparison of predicted and measured values of the specific ultrasound energy consumption  $\vec{E}_s$ 

Both non-linear effects and an interaction effect between the two independent variables are accounted for in this model. In the model, the  $t_{on}$  and the interaction term " $t_{on} \times t_{off}$ " are statistically significant with p-values < 0.05. The term  $t_{off}$  is not statistically so significant in itself, but its

interaction with the term  $t_{on}$  is. In practice, this implies that the effect of  $t_{on}$  on  $\bar{E}_s$  is dependent on the level of  $t_{off}$ . This can be seen in Figure 8, which represents a contour plot showing the effect upon  $\bar{E}_s$  from varying the  $t_{on}$  and  $t_{off}$  within their ranges from 1 to 5 seconds.



**Figure 8.** The contour plot showing the effect of  $t_{on}$  and  $t_{off}$  on the specific ultrasound energy consumption  $\bar{E}_s$ . (blue means lower  $\bar{E}_s$ , red means higher  $\bar{E}_s$ ).

Figure 8 illustrates by colors that  $\bar{E}_s$  depends on the variables  $t_{on}$  and  $t_{off}$ . In the figure, a darker blue means lower  $\bar{E}_s$ , more red means higher  $\bar{E}_s$ . It is seen that the left lower side area (dark blue) represents an area of optimum values for the independent variables. It shows that when  $t_{on}$  is high, increasing  $t_{off}$  become significant for decreasing  $\bar{E}_s$ . The data show a weak saddle structure where  $\bar{E}_s$ have lower values both in the upper right and in the lower left. In the variable range investigated the lowest value range of  $\bar{E}_s$  are found where  $t_{on}$  is 1 s and  $t_{off}$  is 1 to 2 s respectively. The structure of the values in Figure 8 is of such a nature that it would clearly be interesting to investigate even lower  $t_{on}$  than 1 s. This, however, could not be done at this stage because of the operational limitation of the ultrasound device and the experimental rig. Therefore, an optimum condition,  $t_{off}$ was set to 2 s and  $t_{on}$  is 1 s was predicted.

Using equation (9) it was predicted that an optimum condition is  $t_{on} = 1$  s and  $t_{off} = 2$  s where it is predicted that  $\bar{E}_s = 3.99$  MJ/kg CO<sub>2</sub> when CO<sub>2</sub> loading is 0.25 mol/mol. Experimental verification of this, using 3 parallels, gave  $\bar{E}_s = 3.6$  MJ/kg CO<sub>2</sub> which is better than predicted.

#### 4. CO<sub>2</sub> MASS BALANCE AND ENERGY BALANCE ANALYSIS

# 4.1. CO<sub>2</sub> mass Balance

CO<sub>2</sub> mass balance calculation was conducted by comparing the amount of change of CO<sub>2</sub> in the liquid phase and the gas phase during the measuring time. The results are shown in Table 3 showing that the average deviation is 4%, and the maximum deviation is 9% (91% matched) in the measurements. The results manifest that the measurements error is in the acceptable range.

		CO <sub>2</sub> in liquid s	ide	CO	<b>D</b>	
No.	Rich loading	Lean loading	CO <sub>2</sub> stripped out	$A_{\rm CO2}$	CO <sub>2</sub> stripped in	Deviation %
	mol/mol	mol/mol	g	mg/s	g	
Heat only	0.25	0.248	1.64	9.5	1.61	1.3
1	0.25	0.239	8.99	50.5	8.57	4.7
2	0.25	0.238	9.81	54.0	9.18	6.4
3	0.25	0.242	6.54	37.0	6.29	3.9
4	0.25	0.240	8.18	49.2	8.35	2.1
5	0.25	0.245	4.09	24.8	4.23	3.5
6	0.25	0.239	8.99	49.8	8.46	5.9
7	0.25	0.242	6.54	39.7	6.73	3.0
8	0.25	0.244	4.91	26.2	4.45	9.2
9	0.25	0.241	7.36	43.2	7.35	0.2
10	0.25	0.247	2.45	15.3	2.62	6.6
11	0.25	0.242	6.54	40.0	6.79	3.8
Opt.	0.25	0.241	7.36	43.2	7.35	0.2
				/	Average.	4%

Table 3. CO<sub>2</sub> Mass Balance Calculation

Average:

Note: 1) Error cause from loading measurement, and CO<sub>2</sub> flow rate measurement by flow meter.

2) Lean solution samples were taken from the reboiler after treatment, not from the lean tank.

3) Total liquid inventory during the measurement was 3.8 kg.

#### 4.2. Energy Balance

Because the liquid was pre-heated from 23 °C to 110 °C, and will be cooled from 121.4 °C to 23 °C after the measurement, this sensible heat (from 23 to 110 °C) will not be considered in the calculation. To simplify the problem, here the measurement treated by heat only was used as an example, assuming the reboiler as the control element, the energy input into the system  $H_{in}$  is

$$H_{\rm in} = H_{out} \tag{10}$$

When CO<sub>2</sub> stripping without ultrasound (heat treatment only)

$$H_{in} = H_{st} \tag{11}$$

and the energy output from the system  $H_{\text{out}}$  is

$$H_{\rm out} = E_{\rm cw} + E_{\rm less\_preheat} + E_{\rm reb\_heat} + E_{\rm de\_CO2} + E_{\rm loss}$$
(12)

where  $E_{cw}$  is the energy consumption of the condensation of the vapor produced in the reboiler,  $E_{\text{less preheat}}$  is the energy consumption due to the liquid temperature difference between in the reboiler and the inlet of liquid,  $E_{\rm reb heat}$  is the energy consumption due to the temperature difference of the liquid in the reboiler before and after measurement,  $E_{de_{CO2}}$  is the energy consumption due to CO<sub>2</sub> desorption from liquid, the theoretical enthalpy of desorbed CO<sub>2</sub> from 0.25 loading 30% MEA solution at 120 °C is 2.6 MJ/kg CO<sub>2</sub> [19].  $E_{loss}$  is the heat loss of the rig to environment.

The energy balance based on the reboiler as a control element show that the energy consumption is 82% of the energy input. Most of the energy losses were caused by the heat loss of the rig and error of condensed steam collection, considering these factors, 82% energy match is acceptable for this laboratory test. More detail of the calculation can be found in the supporting information.

# 5. COST ESTIMATION

An MEA-based  $CO_2$  capture plant (see the Supporting information) is used as the basis of assessing the cost savings potential of ultrasound implementation. In this case,  $CO_2$  is captured from the flue gas from a CCGT natural gas based power plant. Two cases, one with and one without ultrasound implementation are cost consistently estimated.

# 5.1. Assumptions and Basic Data Calculations

The following sections cover the assumptions used as basis for the cost estimation, both technical and economical.

# 5.1.1. Specific Energy Expected Because of Ultrasound

From Table 2 it is observed that the normalized specific energy for desorption of  $CO_2$  by heat only obtained from measurements in the reboiler rig is reported [17, 18] as an average value 4.2 MJ/kg  $CO_2$ . Based on the value 4.2 MJ/kg  $CO_2$ , an optimized normalized specific energy for desorption of  $CO_2$  by (ultrasound + heat) is 3.6 MJ/kg $CO_2$  in our lab test, and the energy saving is 14% compared to heat only treatment. This saving is conservative because this energy saving does not include the reduction of size of heat exchangers, reboilers and pumps with reduced energy due to the reduction of liquid flow, caused by a leaner regenerated absorbent.

In this case, the stripped CO<sub>2</sub> can be divided into two parts when ultrasound is applied:

- Part 1, in the assumed desorption column, the CO<sub>2</sub> stripping from rich loading to normal lean loading, e.g. from 0.44 to 0.25, Δα=0.19, and the CO<sub>2</sub> is stripped by the vapors from the reboilers produced by steam heat and ultrasound. More vapors (mainly water) will be produced due to the assistance of ultrasound.
- Part 2, in the reboiler, the CO<sub>2</sub> stripping from normal lean loading to extra lean loading, i.e. from 0.25 to 0.20,  $\Delta \alpha$ =0.05, and the energy input is by steam and ultrasound. Assuming an optimum ultrasound application in the reboiler, and;
  - Based on the optimized experiment (in Table 2), 153.3 J/s from ultrasound input, 264.7 J/s from heat input, total stripped CO<sub>2</sub> is (43.2 + 71.5 =) 114.7 mg/s.

- For heat treatment only, the stripped  $CO_2$  is (9.5 + 50 =) 59.5 mg/s.
- Then the extra stripped CO<sub>2</sub> due to ultrasound is (114.7 59.5 =) 55.2 mg/s.
- From the optimized experiment, the normalized total specific energy consumption is  $3.6 \text{ MJ/kg CO}_2$ . In this case, it was assumed that the specific energy consumption of theoretical desorption enthalpy (2.6 MJ/kg CO<sub>2</sub>) [19] is provided by steam ( $H_{st}$ ), and the rest energy consumption (3.6 2.6) MJ/kg CO<sub>2</sub> = 1000 kJ/kg CO<sub>2</sub> is contributed by ultrasound.
- A simulation results of the base case by Aspen Plus show that the CO<sub>2</sub> loading decrease is 0.05 before and after the reboiler (see the supporting information). In this work, the CO<sub>2</sub> stripping rate in the reboiler assisted by ultrasound (43.2 mg/s) is 4.5 times of heat treatment only (9.5 mg/s), indicating that it could achieve  $\Delta \alpha$ =0.05 from 0.25 to 0.20 when assisted by ultrasound in the industrial reboiler.

# **5.1.2.** Implementation of Ultrasound for Cost Estimation Purposes

Possible locations for implementations of ultrasound in the process were described in the introduction. It was observed in the experiments that fewer bubbles are formed from the steam pipe for a while after ultrasound treatment. This means that the ultrasound affects the formation of bubbles on the surface of the steam pipe. Therefore, we suggest that the ultrasound sonotrode should be installed at some distance from the steam pipe.

Based on our experiments, we use the following implementation for cost estimation purposes: Sonotrodes mounted in the main section of the stripper reboiler as illustrated in Figure 9. The sonotrodes will be distributed evenly along the bottom of the reboiler to get maximum effective area for cavitation. The details of ultrasound equipment can be found in the Supporting information.



Figure 9. Illustration of an industrial kettle reboiler, with 5 ultrasound sonotrodes on the bottom.

## 5.1.3. Effect of CO<sub>2</sub> Loading

There is an increased cyclic capacity for the absorbent based on the reduced lean loading to 0.20 mol/mol from 0.44 mol/mol ( $\Delta \alpha_{\text{US}}=0.44 - 0.20=0.24$ ), such that the flow rate can be reduced from the base case ( $\Delta \alpha_{\text{BC}}=0.44 - 0.25=0.19$ ), as follows:

$$\frac{\Delta \alpha_{US} - \Delta \alpha_{BC}}{\Delta \alpha_{US}} = \frac{0.24 - 0.19}{0.24} = 0.21, \qquad i.e.21\% \ reduced \ flow$$

This also implies that 21% of the  $CO_2$  recovered in the process is the extra  $CO_2$  desorbed caused by ultrasound. The solvent flow rate reduction will affect several of the components listed in Table 4. Some of the equipment sizes and energy consumptions are governed by the gas flow, while others are governed by the solvent (MEA) flow. It is the ones governed by the solvent flow that are affected and some more than others. Five components have been identified to undergo the most significant changes including the reboiler, and they are (basic case);

- H-3 Lean/rich solution heat exchanger
- H-4 Lean amine cooler
- Reboiler
- Pump, P-3, the rich solution pump
- Pump, P-4, the lean solution pump

The capacities and reductions of these units because of the reduction of absorbent are calculated as shown in Table 4.

# 5.1.4. Potential Steam Savings

With respect to CCGT, the base case is that there are 15 reboilers and 50.85 kg CO<sub>2</sub>/s capacity in total. Based on the 21% reduction of absorbent flow, the number of reboilers can be reduced to 12. The main CO<sub>2</sub> stripping is from loading 0.44 to 0.25 (part 1),  $\Delta \alpha = 0.19$  and main stripped by the solution vapor produced in the reboiler, and some extra vapor produced because of the application of ultrasound. In part 2, the loading is from 0.25 to 0.20,  $\Delta \alpha = 0.05$ .

Part 1, because the ultrasound could increase the CO<sub>2</sub> stripping in this part, from the experiments ("opti.", "heat") in Table 2, the increase of CO<sub>2</sub> stripping rate is (71.5 - 50)/50=43% due to ultrasound applied, conservative assumption is 22% can be realized in an industrial unit (50% of experimental results). Then the total stripped CO<sub>2</sub> from Part 1 would be

$$50.85 \times \frac{0.19}{0.05 + 0.19} \times (1 + 22\%) = 49.11 \frac{\text{kgCO}_2}{\text{s}}$$

Where  $49.11 \times 22\% = 10.8 \text{ kg/CO}_2$  is contributed by introduction of ultrasound for the part 1.

- Part 2, the total CO<sub>2</sub> stripped is (50.85 49.11=) 1.74 kg CO<sub>2</sub>/s
  - Specific energy consumption from ultrasound is 1000 kJ/kg CO<sub>2</sub>
  - The total energy input from ultrasound equipment (of 12 reboilers) is

$$1000 \frac{\text{kJ}}{\text{kg CO}_2} \times \left(1.74 \frac{\text{kg CO}_2}{s} \times \frac{55.2}{114.7}\right) = 838 \, kW$$

Where 55.2 mg/s is the CO<sub>2</sub> stripping due to ultrasound, 114.7 mg/s is the total CO<sub>2</sub> stripping by heat and ultrasound in the reboiler.

Assuming that one ultrasound sonotrode unit can supply 16 kW, we need

$$\frac{838 \, kW}{16 \, kW \times 12} \approx 5 \, (ultrasound \, units \, per \, reboiler)$$

Commercial ultrasound devices have been identified and available for industrial purposes on a medium and large scale.

It is assumed 2.6 MJ/kg  $CO_2$  of the 3.6 MJ/kg  $CO_2$  (the normalized total specific energy consumption) is the heat needed for theoretic desorption. The steam needed for the normal case is conservatively 4.2 MJ/kg  $CO_2$ . Because the heat transfer in the reboilers cannot be 100%, the steam reduction is

$$(4.2 - \frac{2.6}{88\%})/4.2 = 30\%$$

Where 88% is the heat efficiency in our lab rig.

# 5.2. Changes in Equipment

In Table 4, the base case equipment list is shown, note that  $CO_2$  compression is not included. The flowsheet is shown in Supporting information. The heat exchangers area, reboilers and pump sizes are reduced because of the reduction of absorbent circulation. Ultrasound equipment is added in the process.

Tag nr.	Description	Unit	Size	Amount	Change	New size	New amount
H-1	Cooling water cooler	m²	5 435	6			
H-2	Wash water cooler	m²	7 350	8			
H-3	Lean/rich solution HE	m²	33 865	34	7 044	26 821	27
H-4	Lean solution cooler	m²	1 520	2	316	1 204	2
H-5	Stripper condenser	m²	1 665	2			6
H-6	Stripper reboiler	m²	14 160	15	2 945	11 215	12
V-1	DCC	m³	7 270	2			
(V-1)	Packing DCC	m³	1 450	2			
V-2	Absorber shell	m³	16 290	2			
(V-2)	Packing (absorber)	m³	6 790	2			
V-3	Water wash (absorber)	m³	1 250	2			
(V-3)	Packing (water wash)	m³	940	2			
V-4	Reclaimer	m²	550	1			
V-5	Stripper shell	m³	1 270	1			
(V-5)	Packing (stripper)	m³	510	1			
V-6	Separator	m³	16	1			
V-7	Lean solvent tank	m³	1 180	2			
V-8	Amine supply tank	m³	200	1			
V-9	Amine/chemicals mixing tank	m³	8	_1			
V-10	Amine sump	m³	40	1			
P-1	DCC water pump	kW	320	1			
P-2	Wash water circ. pump	kW	450	1			
P-3	Rich solution pump	kW	890	2	185	705	2
P-4	Lean solution pump	kW	890	2	185	705	2
P-5	Condenser return pump	kW	22	1			
P-6	Amine storage tank pump	kW	22	1			
P-7	Water injection pump	kW	22	1			
P-8	Amine Fill pump	kW	12	1			
P-9	Water makeup pump	kW	22	1			
P-10	Condensate pump	kW	22	1			
P-11	Amine Sump pump	kW	22	1			
K-1	Flue gas fan	kW	5 075	3			
F-1	Filter package	-		1			
X-2	Soda ash package	-		1			
	ultrasound equipment	kW				838	60

Table 4.	Base	case ec	quipment	list,	with	changes	due t	o ultrasound	apr	olication.
						<i>u</i>				

# 5.3. Results of Cost Estimation

The cost estimates without (base case) and with the ultrasound have been done using the same flowsheet and equipment list, see Table 4. In the currently chosen configuration, five ultrasound sonotrodes at 16 kW is implemented in each kettle reboiler, as illustrated in Figure 9. The installation cost of the ultrasound sonotrodes was included in the kettle reboiler costs, resulting in an increased unit cost and installation factor compared to base case. The operational cost (electricity) was added separately.

The assumptions in the cost estimates were kept the same for both estimates. The two most important ones are related to energy cost; steam and electricity. When implementing ultrasound, a

part of the steam consumption is replaced by electricity. Therefore, the cost of these elements is important, and in the current estimates their prices are as follows;

- Steam, 21.3 EUR/t
- Electricity, 0.05 EUR/kWh

The CAPEX of one ultrasound sonotrode unit was based on a quote from a supplier, 2200 EUR for one 2 kW sonotrode. The cost of one 16 kW sonotrode was estimated 8850 EUR  $(=2200\times(16/2)^{0.65})$ .

The cost of the ultrasound equipment is included in the reboiler cost at an increased unit cost and installation cost. This is likely to be the case for a new build where the ultrasound sonotrodes will be a highly integrated part of the reboiler and delivered as a package.

The results, capture cost only,  $CO_2$  compression is not included, are shown in Table 5. The results are divided into CAPEX, OPEX and total capture cost, and the percentage improvement is included for each. The result showed a slight increase in CAPEX, with a more pronounced reduction in OPEX, with current assumptions.

Table 5. The	e results from the cost es	stimation, reference y	year 2018.
	CAPEX	OPEX	Total capture cost
	EUR/t CO <sub>2</sub>	EUR/t CO <sub>2</sub>	EUR/t CO <sub>2</sub>
Base case	11.2	47.6	58.8
Process with ultrasound	10.8	36.6	47.4
Cost saving	4%	23%	19%

As briefly discussed above, the results are dependent on the cost of utilities. Low steam cost and high electricity costs will favor the base case, while the opposite will favor the ultrasound modified base case. It should also be mentioned that estimates for the modified base case are conservative, primarily regarding the electricity consumption, but also likely concerning ultrasound sonotrodes' CAPEX.

It is noted that the cost estimation is sensitive to the  $CO_2$  loading of the lean solution, i.e., leaner solution leads to a higher capacity of the solution and then less MEA inventory.

# 6. CONCLUSIONS AND RECOMMENDATIONS

The use of ultrasound to improve the desorption of  $CO_2$  from lean loaded solution was investigated covering a typical industrial case of a reboiler pressure of 1.0 barg. A test with varying ultrasound exposure times was performed from which the optimum times of  $t_{on} = 1$  s and  $t_{off} = 2$  s was found in current lab-rig. The results show that the enhancement of  $CO_2$  stripping by ultrasound is significant and a 300% improvement is obtained, indicating that the mass transfer can be intensified by ultrasound. This energy saving per kg of  $CO_2$  is a direct consequence of the larger

amount of CO<sub>2</sub> produced using ultrasound. To be able to compare with the typical MEA-based CO<sub>2</sub> plant, a normalized specific energy was defined based on industrial case (4.2 kJ/kg CO<sub>2</sub>), it was deduced that the specific energy consumption (normalized) in an industrial reboiler is 3.6 MJ/kg CO<sub>2</sub>, and the energy saving reached 14% when the extra vapor (mainly water) by ultrasound that enters an assumed desorption column for further CO<sub>2</sub> stripping is considered.

Cost estimations have been conducted by using Aspen Plus V9.0 and Aspen In-plant Cost Estimator for the industrial cases with/without the assistance of ultrasound. Total capture cost including CAPEX and OPEX is 60.2 EUR/t CO<sub>2</sub> and cost saving is 19% when the CO<sub>2</sub> loading in the MEA solution can be decreased to 0.20 mol/mol from 0.44 mol/mol assisted by ultrasound.

# ASSOCIATED CONTENT

#### Supporting Information

The Energy balance calculations and the flow sheet of  $CO_2$  capture plant for CCGT natural gas based power plant can be found in the supporting information.

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#### ACKNOWLEDGMENTS

This work was funded by Shell Technology Norway AS and Norwegian Research Council (No.235055).

## NOMENCLATURE

Abbreviations

MEA = monoethanolamine

CCGT= Combined Cycle Gas Turbine

Parameters and Variables

 $A_{\rm CO2}$ , is the CO<sub>2</sub> stripping rate in the reboiler, mg/s

 $A_{\rm CO2,H}$ , is the CO<sub>2</sub> stripping rate by heat only, mg/s

 $A_{\text{CO2,US}}$ , is the CO<sub>2</sub> stripping rate by heat + ultrasound, mg/s,  $A_{\text{CO2,US}} = A_{\text{CO2}} - A_{\text{CO2,H}}$ 

 $A_{CO2,v}$ , is the CO<sub>2</sub> stripping rate of by the vapor in the assumed desorption column, mg/s

 $A_{\rm CO2,v,H}$  is the CO<sub>2</sub> stripping rate by the vapor in the assumed desorption column when heat treatment only, mg/s

 $A_{\text{CO2,v,US}}$ , is the stripping rate of CO<sub>2</sub> further stripped in the assumed desorption column by vapor when ultrasound is introduced, mg/s

 $E_{\rm cw}$ , is the energy consumption of the vapor produced in the reboiler condensed to liquid state, MJ/min

 $E_{de_{CO2}}$ , is the energy consumption due to CO<sub>2</sub> desorption from liquid, MJ/min

 $E_{\text{less_preheat}}$ , is the energy consumption due to the liquid temperature difference between in the reboiler and the inlet of liquid, MJ/min

 $E_{\rm loss}$ , is the heat loss of the rig in the measurement, MJ/min

 $E_{\text{reb}\_heat}$ , is the energy consumption due to the temperature difference of the liquid in the reboiler before and after measurement, MJ/min

 $H_{\rm in}$ , is the energy input into the control element, MJ/min

 $H_{\rm out}$ , is the energy output from the control element, MJ/min

 $H_{\rm st}$ , is the energy input by steam heat, MJ/min

H<sub>US</sub>, is the ultrasound energy input, MJ/min

 $E_{\rm s}$ , is the specific energy consumption in reboiler, MJ/kg CO<sub>2</sub>

 $E_{\rm s,H}$ , is the specific energy consumption when CO<sub>2</sub> stripping only by heat in the experiment, MJ/kg CO<sub>2</sub>

 $E_{s,US}$ , is the specific energy consumption when CO<sub>2</sub> stripping assisted by ultrasound in the experiment, MJ/kg CO<sub>2</sub>

 $\bar{E}_{s}$ , is the normalized specific energy consumption, MJ/kg CO<sub>2</sub>

 $\bar{E}_{s,H}$ , is the typical specific energy consumption of a MEA plant, i.e. 4.2 MJ/kg CO<sub>2</sub>

 $\bar{E}_{s,US}$ , is the normalized specific energy consumption when CO<sub>2</sub> stripping assisted by ultrasound, MJ/kg CO<sub>2</sub>

 $t_{\rm on}$ , on-time of ultrasound in a period, s

 $t_{\rm off}$ , off-time of ultrasound in a period, s

 $W_{\rm v,H}$ , is the weight of vapor produced by heat only in unit time, mg/s

 $W_{\rm v,US}$ , is the weight of vapor produced in unit time when ultrasound is introduced, mg/s

Greek Symbols

 $\alpha$ , is the CO<sub>2</sub> loading in MEA aq. solution, mol CO<sub>2</sub>/mol MEA

 $\alpha_{rich}$ , is the CO<sub>2</sub> loading in rich MEA aq. solution, mol CO<sub>2</sub>/mol MEA

 $\alpha_{lean}$ , is the CO<sub>2</sub> loading in lean MEA aq. solution, mol CO<sub>2</sub>/mol MEA

 $\Delta \alpha_{BC}$ , is the CO<sub>2</sub> loading change MEA aq. solution from rich to lean solution of the base case, mol CO<sub>2</sub>/mol MEA

 $\Delta \alpha_{US}$  is the CO<sub>2</sub> loading change MEA aq. solution from rich to lean solution of the base case when ultrasound applied, mol CO<sub>2</sub>/mol MEA

 $\gamma$ , is the energy efficiency of steam heat input, 88% measured from a blank experiment (water in the reboiler) at room pressure and the steam temperature is set to 130 °C

 $\lambda$ , is the improvement of CO<sub>2</sub> stripping rate, %

 $\varphi$ , is the "on" time fraction of ultrasound in a period, %

 $\eta$ , is the energy saving due to using ultrasound in the reboiler process, %

 $\bar{\eta}$ , is the normalized energy saving for a process, %

# REFERENCES

[1] Mason TJ. Advances in Sonochemistry: JAI Press, 1999.

[2] Brennen CE. Cavitation and bubble dynamics: Cambridge University Press, 2013.

[3] Pilli S, Bhunia P, Yan S, LeBlanc R, Tyagi R, Surampalli R. Ultrasonic pretreatment of sludge: a review. Ultrasonics sonochemistry. 2011;18(1):1-18.

[4] Tanaka K, Fujiwara T, Okawa H, Kato T, Sugawara K. Ultrasound irradiation for desorption of carbon dioxide gas from aqueous solutions of monoethanolamine. Japanese Journal of Applied Physics. 2014;53(7S):07KE14.

[5] Zhang J, Qiao Y, Agar DW. Intensification of low temperature thermomorphic biphasic amine solvent regeneration for CO 2 capture. Chemical Engineering Research and Design. 2012;90(6):743-9.

[6] Gantert S, Möller D. Ultrasonic desorption of CO2–a new technology to save energy and prevent solvent degradation. Chemical Engineering & Technology. 2012;35(3):576-8.

[7] Xue J, Shen B, Du S, Lan X. Study on desorbing sulfur dioxide from citrate solution by ultrasonification. Chinese Journal of Chemical Engineering. 2007;15(4):486-91.

[8] Schueller BS, Yang RT. Ultrasound enhanced adsorption and desorption of phenol on activated carbon and polymeric resin. Industrial & engineering chemistry research. 2001;40(22):4912-8.

[9] Leong T, Ashokkumar M, Kentish S. The Growth of Bubbles in an Acoustic Field by Rectified Diffusion.

Handbook of Ultrasonics and Sonochemistry. Singapore: Springer Singapore; 2016. p. 69-98.

[10] Peregrine D. The Acoustic Bubble. Journal of Fluid Mechanics. 1994;272:407-8.

[11] Doinikov AA. Bjerknes forces and translational bubble dynamics. Bubble and particle dynamics in acoustic fields: modern trends and applications. 2005;661:95-143.

[12] Ying J, Eimer DA, Mathisen A, Sørensen H, Haugen HA. Intensification of CO2 Stripping from Amine Solutions by Ultrasonic. Energy Procedia. 2014;63:781 – 6.

[13] Ying J, Eimer DA, Haugen HA. Ultrasound to assist desorption of CO2 from loaded monoethanolamine solutions. 14th Meeting of the European Society of Sonochemistry. Avignon, France2014. p. 235-6.

[14] Ying J, Haverkort J, Eimer DA, Haugen HA. Ultrasound enhanced CO2 Stripping from Lean MEA Solution at Pressures from 1 to 2.5 bar (a). Energy Procedia. 2017;114:139-48.

[15] Ying J, Eimer DA, Brakstad F, Haugen HA. Ultrasound intensified CO2 desorption from pressurized loaded monoethanolamine solutions. I. parameters investigation and modelling. Energy. 2018;163:168-79.

[16] Box GE, Hunter JS, Hunter WG. Statistics for experimenters: design, innovation, and discovery: Wiley-Interscience New York, 2005.

[17] Rochelle G, Chen E, Freeman S, Van Wagener D, Xu Q, Voice A. Aqueous piperazine as the new standard for CO2 capture technology. Chemical engineering journal. 2011;171(3):725-33.

[18] Chapel DG, Mariz CL, Ernest J. Recovery of CO2 from flue gases: commercial trends. Conference Recovery of CO2 from flue gases: commercial trends, vol. 4. Saskatchewan Canada.

[19] Kim I, Svendsen HF. Heat of absorption of carbon dioxide (CO2) in monoethanolamine (MEA) and 2-(aminoethyl) ethanolamine (AEEA) solutions. Industrial & engineering chemistry research. 2007;46(17):5803-9.

Highlights:

- 1. Ultrasound is introduced to intensify CO<sub>2</sub> stripping from loaded amine solutions.
- 2. On/off time of ultrasound is optimized in simulated industrial conditions.
- 3. Energy saving 14% and cost saving 19% are achieved by ultrasonic assistance.