

1 **Are fluorescence-based chlorophyll quantification methods suitable for**
2 **algae toxicity assessment of carbon nanomaterials?**

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9 **Are fluorescence-based chlorophyll quantification methods suitable for algae**
10 **toxicity assessment of carbon nanomaterials?**

11
12 Using a multi-walled carbon nanotube (MWCNT) and graphene oxide (GO) as representative test
13 materials, we evaluated the applicability of *in vivo* and *in vitro* chlorophyll-a (Chl-a) fluorescence
14 quantification methods, which are used in standard algae ecotoxicity tests such as OECD 201 and ISO
15 8692. *In vivo* quantification of Chl-a from *Raphidocelis subcapitata* indicated a significant reduction in
16 Chl-a fluorescence in the presence of MWCNTs due to shading, but a significant autofluorescence from
17 GO that caused an overestimation of Chl-a concentration. *In vitro* Chl-a quantification methods
18 employing a modified acetone and an ethanol extraction protocol reduced the influence of shading and
19 autofluorescence, but both resulted in a significant loss of fluorescence signal in the presence of 100 mg
20 L⁻¹ MWCNTS (99-100%) and GO (21-52%). Chl-a reduction was dose dependent for both tested CNM
21 materials, but effects were more pronounced for MWCNT, which caused a significant fluorescence
22 reduction (16±0.3%) already at 1 mg L⁻¹. Further study of the CNM-algae-Chl-a interaction processes
23 revealed that CNM can not only interact with live algae, but also efficiently adsorb extracted Chl-a. Our
24 results showed that within 10 min, 95-100% of Chl-a extracted from two algae concentrations were
25 adsorbed to MWCNT, while 35-60% of Chl-a was adsorbed to the GO. This study shows that Chl-a
26 quantification by fluorescence determination is not a suitable method for ecotoxicity testing of CNM.
27 However, a quick screening test for individual MNMs is recommended to determine if Chl-a adsorption
28 is a significant process prior to selection of a quantification method.

29
30 Key words: Standard tests; adsorption; carbon nanotubes; graphene oxide;

31 **Introduction**

32 Many of the existing standard toxicity test guidelines (TGs) have been found to exhibit
33 limitations when applied to assessment of manufactured nanomaterials (MNM)s (Sørensen *et al.*, 2015). As a result, there has been significant effort within the Organization for Economic
34 Cooperation and Development (OECD), the International Standardization Organization (ISO)
35 and large-scale research projects (e.g. EU FP7 projects Marina and NANoREG) to evaluate
36 the applicability of existing TGs for MNMs. In 2009, the OECD Working Party on
37 Manufactured Nanomaterials (WPMN) assessed the limitations of existing OECD TGs for
38 MNMs (OECD, 2009), and published the results of a seven year testing programme as a
39 series of dossiers in 2015. The general conclusion of OECD was that most TGs are considered
40 suitable for the testing of MNMs, but common issues of harmonised dispersion preparation,
41 sedimentation during exposure and dosimetry were highlighted. These issues have been the
42 subject of individual studies (Hartmann *et al.*, 2010, Hund-Rinke *et al.*, 2010, Hartmann *et al.*,
43 2015) and a critical review of the dossiers has highlighted significant limitations with the
44 conclusions drawn by OECD (Hansen *et al.*, 2017). In some cases, modifications were
45 proposed for improving the relevancy of specific TGs towards MNMs.
46

47

48 In 2012, a guidance document addressed general issues such as sample preparation
49 and dosimetry, but lacked specific focus on individual TGs (OECD, 2012). Despite further
50 work, there is currently no official compilation of proposed nano-specific test modifications
51 on OECD TGs for ecotoxicity, therefore the adaption of the test guidelines is hampered
52 (Hund-Rinke *et al.*, 2016). Most recently, the EU FP7 project Marina has provided an
53 overview of the progress on eight OECD ecotoxicity TGs, summarising the difficulties
54 identified with individual TGs for MNMs and the proposed modifications.

55

56 OECD Guideline 201 (Freshwater Alga and Cyanobacteria, Growth Inhibition Test)
57 (OECD, 2011) and ISO 8692:2004 (Water quality–freshwater algal growth inhibition test
58 with *Scenedesmus subspicatus* and *Selenastrum capricornutum*) (ISO, 2004) are two of the
59 most commonly used standard aquatic ecotoxicity tests. The tests are quick and cost effective
60 to perform, reliable and reproducible, employ organisms which are easy to maintain in
61 permanent culture, and present the user with few ethical considerations compared to other
62 species. The test is internationally recognised, and freshwater algae toxicity data are available
63 for many chemicals and metals. However, significant variability in EC₅₀ values for MNM
64 effects on algal growth is reported, highlighting the importance of test reliability and
65 reproducibility for interpretability (Menard *et al.*, 2011).

66
67 OECD 201 is considered to have limited issues with respect to sedimentation and
68 dosimetry during exposure, due to samples being shaken during the incubation period (Hund-
69 Rinke *et al.*, 2016). However, two key limitations regarding the applicability of OECD 201 to
70 MNMs have been identified. The first is that the chelating agent EDTA, a component of the
71 test medium, can interfere with metallic MNMs causing iron precipitation and resulting in
72 unfavourable growth conditions (Kadar *et al.*, 2012, Hund-Rinke *et al.*, 2016, Mouchet *et al.*,
73 2016, Zhang *et al.*, 2016). It is now recommended that an EDTA-free algal medium should be
74 considered when testing metal MNMs (Hund-Rinke *et al.*, 2016). The second is accuracy in
75 determination of algal growth. Biomass determination through dry weight is not applicable in
76 the presence of particles and interference was reported in both cell counting and *in vivo*
77 measurement of chlorophyll-a (Chl-a) concentrations by fluorescence (Handy *et al.*, 2012a,
78 Handy *et al.*, 2012b, Hartmann *et al.*, 2012, Hund-Rinke *et al.*, 2016). Cell counting in a
79 haemocytometer is very laborious, has a large variance and may not truly reflect algal
80 biomass if the mean cell size changes in response to a toxicant or other conditions (Hund-

81 Rinke *et al.*, 2016). In the case of *in vivo* fluorescence Chl-a measurements, MNMs can cause
82 shading of the algae fluorescent signal, which reduces the response, underestimating the Chl-a
83 concentration and overestimating the toxicity of the MNM being studied (Handy *et al.*, 2012a,
84 Handy *et al.*, 2012b, Hartmann *et al.*, 2012, Sørensen *et al.*, 2015, Hund-Rinke *et al.*, 2016,
85 Sørensen *et al.*, 2016). The term shading can refer to either (i) blocking of the light from the
86 source to the algal cells, or (ii) blocking of the fluorescence signal from the algal cells to the
87 detector. The main approach suggested for overcoming the issue of fluorescence signal
88 shading is the extraction of the Chl-a from the algal cells and removal of the MNMs prior to
89 *in vitro* fluorescence measurement of the algal extract (Hartmann *et al.*, 2012, Hund-Rinke *et*
90 *al.*, 2016). Methods proposed include the use of acetone extraction (Mayer *et al.*, 1997,
91 Sørensen *et al.*, 2016), ethanol extraction and filtration (ISO 10260) (ISO, 1992), and acetone
92 extraction combined with MNM flocculation and sedimentation (no filtration) using locust
93 bean gum (Cerrillo *et al.*, 2016, Hund-Rinke *et al.*, 2016).

94
95 Although these studies used different reference MNMs (TiO₂, Ag and Au) (Hartmann
96 *et al.*, 2012, Hund-Rinke *et al.*, 2016), no carbon-based MNMs (CNMs) were included.
97 Carbon nanotubes (CNTs) and graphene family nanomaterials (GFNs) represent a large group
98 of CNMs with increasing production volumes. CNTs and GFNs exhibit unique
99 physicochemical properties that differ quite significantly from spherical metal and metal
100 oxide MNMs and have implications for their environmental fate and effects (Jackson *et al.*,
101 2013, Zhao *et al.*, 2014, Hu *et al.*, 2016). One feature in particular, is the high adsorption
102 capacity of CNMs for organic chemicals (Chen *et al.*, 2007, Kah *et al.*, 2014, Glomstad *et al.*,
103 2016, Zindler *et al.*, 2016). The propensity for CNMs to adsorb organic compounds raises the
104 issue of Chl-a adsorption in algae ecotoxicity studies employing extraction methods.

105

106 In the current study, we evaluate the applicability of different fluorescence-based Chl-
107 a quantification methods for the assessment of CNMs using a multi-walled CNT (MWCNT)
108 and a graphene oxide (GO) as test materials. The suitability of *in vivo* fluorescence
109 approaches are evaluated using OECD Guideline 201 (OECD, 2011). *In vitro* fluorescence
110 approaches are evaluated using a Chl-a extraction method described in the ISO 10260
111 guideline (ISO, 1992), and also by applying a modified acetone extraction and flocculation
112 method developed in the EU FP7 project Marina (Hund-Rinke *et al.*, 2016) and benchmarked
113 in the EU FP7 project NANoREG (Cerrillo *et al.*, 2016).

114

115 **Methods**

116 ***Materials***

117 The MWCNTs were supplied by the Joint Research Centre and is known widely as NM-400.
118 The GO was supplied by Abalonyx AS (Oslo, Norway). Nanomaterial stock dispersions of
119 2560 mg L⁻¹ were prepared by probe sonication (Branson Ultrasonics Sonifier S-450; Branson
120 Ultrasonics, USA) at 15 % amplitude for 13 minutes and 45 seconds. This method delivers an
121 acoustic delivered power of 7.35±0.05 Watt and an accumulated dose of 7056±103 J to the
122 dispersion, which was determined using the calorimetric described by Taurozzi *et al.* as a
123 basis (Taurozzi *et al.*, 2012). Stock dispersions of GO were prepared in MilliQ water. Due to
124 the hydrophobicity of the MWCNT, the dispersion procedure was enhanced through the
125 inclusion of a pre-wetting step with 96% ethanol and the addition of 20 mg L⁻¹ natural organic
126 matter (NOM, Suwannee river). Working dispersions were prepared by diluting the stock
127 dispersions either in OECD algae growth media TG201 (for GO exposures) or TG201
128 containing 10 mg L⁻¹ NOM (TG201+NOM; for MWCNT exposures). NOM was dissolved in
129 MilliQ or TG201 by magnetic stirring for 24 h followed by repeated filtration (0.2 µm

130 Nalgene filter unit, Thermo Fisher Scientific Inc., USA) to remove undissolved material.

131

132 ***Nanomaterial Characterisation***

133 SEM was performed with an Hitachi S-5500 ultra high resolution SEM equipped with bright

134 field and dark field STEM detectors and a Bruker Quantax EDS system. The MWCNT

135 samples were dispersed in isopropanol and prepared on lacey carbon TEM grids. GO was

136 dispersed in high purity water and similarly prepared on a lacey carbon grid. Prior to imaging,

137 all samples were cleaned of volatile organic contaminants using a UV ozone cleaner. An

138 acceleration voltage of 30kV was used for images. The hydrodynamic diameter (HDD) and

139 zeta potential of MWCNT and GO were determined with a Zetasizer ZS (Malvern

140 Instruments, UK) in stock dispersions and exposure dispersions (TG201 and TG201+NOM)

141 prepared at a concentration of 100 mg L⁻¹. It is acknowledged that the usefulness of dynamic

142 light scattering is limited for MNMs with high aspect ratios and high polydispersity (e.g.

143 CNTs).

144

145 ***Algae culture***

146 The algae used for the experiments in this study derived from a continuous in-house

147 *Raphidocelis subcapitata* (formerly known as *Pseudokirchneriella subcapitata*) culture. The

148 algae culture is maintained at 60 – 120 µE/m²/s at a constant temperature of 21±2 °C. Regular

149 monitoring confirms that the culture fulfils the recommended quality criteria regarding

150 growth rate and sensitivity for ecotoxicological testing as specified in ISO 8692:2004 (ISO

151 2004).

152

153 ***Chl-a determination***

154 In this study, we determined Chl-a as surrogate for total Chl. All Chl-a measurements were
155 performed using a fluorimeter (TD700, Turner design, USA) with an optical kit (10-037R;
156 Turner design, USA) suited for *in vivo* and extracted Chl-a measurements (excitation 340-500
157 nm; emission >665 nm).

158

159 ***In vivo Chlorophyll-a determination***

160 In order to investigate the feasibility of *in vivo* Chl-a determination in algae exposed to CNM,
161 MWCNT and GO were added to algae dispersions and the fluorescence signal determined.
162 Algae were diluted in either TG201 (for GO exposures) or TG201+ NOM (MWCNT
163 exposures) to result in fluorescence values of approximately 25 AU (arbitrary units), which
164 corresponds to approximately 25000 cells mL⁻¹ (See Supplementary Information section for
165 more details). Following dilution, MWCNT and GO were added to the algae to reach a final
166 concentration of 100 mg L⁻¹. Algae-CNM dispersions were carefully shaken and left standing
167 still in the dark for 10 minutes to enable algae-CNM interactions, but prevent any
168 toxicological response, before measuring fluorescence as described above. Algae prepared at
169 similarly concentrations in TG201 and TG201+NOM, respectively were used as controls.
170 Control and CNM exposure groups were measured in triplicates (*n*=3).

171

172 ***In vitro Chlorophyll-a determination***

173 *In vitro* Chl-a determination was evaluated using two different Chl-a extraction methods in
174 order to investigate the effect of the extraction solvent on potential CNM interference.
175 Experiments were conducted at algal concentrations resulting in an approximate fluorescence
176 of 500 AU, which corresponds to approximately 500000 cells mL⁻¹ (See Supplementary
177 Information section for more details). These concentrations were based on typical values

178 found in previously conducted OECD 201 growth inhibition tests during the middle-end
179 phase of the test. The CNMs were added to the algae to reach a concentration of 100 mg L⁻¹,
180 selected based on the CLP hazard classification limit. The samples were carefully agitated,
181 left standing for 10 min and extracted thereafter with the two methods described below. As
182 controls, only algae in the respective growth media (TG201 or TG201+NOM) were extracted.
183 Each condition was tested in triplicates ($n=3$). As first method we applied an acetone
184 extraction of Chl-a as described in by Hund-Rinke et al., (Hund-Rinke *et al.*, 2016) and
185 Cerrillo et al., (Cerrillo *et al.*, 2016). In brief, 2 mL of the exposure dispersions were
186 transferred into 8.8 mL of acetone containing lotus bean gum (LBG), mixed by immersion
187 and placed in the dark for 24 h before determining the fluorescence as described above. This
188 method was adapted to testing MNMs using LBG to increase the flocculation and
189 sedimentation of MNMs so that an MNM-free supernatant containing the extracted Chl-a can
190 be isolated and analysed.

191

192 The second method investigated for extracting Chl-a for *in vitro* quantification was the
193 ISO 10260 '*Measurement of biochemical Parameters - Spectrometric determination of the*
194 *chlorophyll-a concentration*', using filtration (GF-F; Whatman, UK) and heated Chl-a
195 extraction in 90% ethanol. The extract was allowed to cool to room temperature and filtered
196 once more in order to remove particulate matter before measuring Chl-a fluorometrically as
197 described above.

198

199 ***Dose dependency***

200 Dose dependent impacts of CNMs on Chl-a quantification were studied using the modified
201 acetone extraction method described above. MWCNT and GO were added to a high algae
202 concentration (fluorescence approximately 600 AU) to reach the following exposure

203 concentrations: 100, 50, 10, 1, 0.1 and 0 mg L⁻¹. The exposures were carefully shaken and left
204 standing still for approximately 10 minutes. To obtain homogeneous samples, the exposure
205 dispersions were agitated and a 2 mL sample was transferred into 8.8 mL of acetone
206 containing LBG, mixed by immersion and placed in the dark for 24 h. Chl-a fluorescence was
207 determined fluorometrically as described above. Each condition was tested in triplicates
208 ($n=3$).

209

210 ***Extraction point and time dependence***

211 In this experiment we wanted to test if CNM interference derives from adsorption and
212 following removal of algae through settling, or if CNMs can also interact with extracted Chl-
213 a. Therefore, 100 mgL⁻¹ MWCNT and GO were added to two algae concentrations: low
214 (fluorescence approximately 15 AU) and high (fluorescence approximately 500 AU). The
215 CNMs were added (i) to the algae 10 min before extraction, (ii) directly after the Chl-a
216 extraction with acetone (as described above), and iii) after 24 h after Chl-a extraction, 10
217 minutes before fluorescence measurements. Due to potential interference of CNMs present in
218 the dispersions, the later samples (iii) were filtered either through GF-F filters (Whatman,
219 UK; MWCNT) or through 0.1 µm Omnipore PTFE filter (MerkMillipore Ltd, Ireland; GO).
220 Following, the fluorescence was determined as described above. All conditions were tested as
221 triplicates ($n=3$).

222

223 ***Microscopy***

224 In order to visualise the interaction of CNM with live *R. subcapitata*, samples of high algae
225 concentrations (600-700 AU) were mixed by agitation with CNMs to reach concentrations of
226 10 or 50 mg L⁻¹ CNMs. Samples were inspected with a fluorescent microscope (Nikon

227 Eclipse 90i; Nikon, Japan) using a 10 and 20x objective and B-2A filter (optimised for blue
228 excitation with a wide excitation bandpass). Further, the adsorption of Chl-a onto MWCNT
229 was visualised. A highly concentrated Chl-a suspension was extracted from an enriched *R.*
230 *subcapitata* culture using the previously described ISO method. The extracted Chl-a was
231 subsequently added to 5 mL of 100 mg L⁻¹ MWCNT and carefully shaken for 2 minutes and
232 left standing in the dark for approximately 10 minutes. As background control dispersions
233 containing only 100 mgL⁻¹ MWCNT were used. The samples were filtered through 0.1 µm
234 Omnipore PTFE filter (MerkMillipore Ltd, Ireland; GO), and the filters placed on microscopy
235 slides. Images of MWCNT were taken with a Nikon Eclipse fluorescent microscope (Nikon,
236 Japan) using a 40x objective and B-2A filter. Chl-a was visualised using fluorescence
237 applying high exposure times (1.5 s) and high contrast mode.

238

239 ***Statistics***

240 Data analyses were performed with GraphPad Prism 7 (GraphPad Software Inc., USA). Data
241 sets were analysed for normality (Shapiro-Wilk normality test). Significant differences
242 between treatments groups were analysed using either t-test (two groups) or one way
243 ANOVA (multiple groups).

244

245 **Results**

246 ***CNM Characterisation***

247 The physicochemical properties of the CNMs used in the study are summarised in Table 1.
248 MWCNT NM-400 was supplied by the JRC repository and sub-samples of material from the
249 same batch has been extensively characterised previously (Rasmussen *et al.*, 2014).
250 Therefore, existing data are presented for size (thickness and length) and specific surface area

251 (SSA). The SSA of the MWCNT was reported as $189.3 \pm 8.1 \text{ m}^2 \text{ g}^{-1}$ using SAXS analysis and
252 $254 \text{ m}^2 \text{ g}^{-1}$ using BET method. The MWCNT had a reported average length of $846 \pm 446 \text{ nm}$,
253 and widths of $16.2 \pm 3.5 \text{ nm}$ and $11 \pm 3 \text{ nm}$ determined independently by two different
254 laboratories (Rasmussen *et al.*, 2014). In house acquired STEM images of MWCNT NM-400
255 showed a similar size range (Figure 1A and B). The GO sheets were relatively heterogeneous,
256 ranging from $<100 \text{ nm}$ to $1 \mu\text{m}$ in size (Figure 1C). The CNMs were also characterised in
257 dispersion form, both as stock dispersions and in the standard algae toxicity media (TG201)
258 used in the studies. Average particle size distribution (Z-ave) for MWCNTs was 3540 ± 727
259 nm in the stock dispersion and $2480 \pm 1450 \text{ nm}$ at 100 mg L^{-1} in TG201+NOM (Table 1). The
260 large Z-ave value and high standard deviation is consistent with high aspect ratio MNMs
261 analysed by dynamic light scattering. The zeta potential values were $0.8 \pm 1.6 \text{ mV}$ and -20 ± 2.8
262 mV , respectively, confirming unstable dispersions of MWCNT. Dispersions of GO were
263 observed to be relatively stable and had a hydrodynamic diameter (HDD) of $245 \pm 14.6 \text{ nm}$ in
264 the stock dispersion and slightly smaller size of $214 \pm 7.22 \text{ nm}$ in the algae growth medium
265 TG201 at a concentration of 100 mg L^{-1} . The zeta potential values were $-41 \pm 0.14 \text{ mV}$ and -
266 $28 \pm 0.25 \text{ mV}$, respectively (Table 1).

267

268 ***In vivo Chlorophyll-a determination***

269 *In vivo* Chl-a fluorescence measurements revealed a high level of interference in the presence
270 of both tested CNMs (Figure 2). Relative to the control samples, Chl-a fluorescence was
271 significantly reduced for MWCNTs through shading of the algal cell fluorescent response,
272 being measured as a significant ($P < 0.001$) net loss of fluorescence in the system (Figure 2).
273 Furthermore, the algae were observed to adsorb or attach significantly to the MWCNT
274 aggregates (Figure 3A). In contrast, the measured fluorescence of the GO at both high and
275 low concentrations was significantly ($P < 0.001$) increased relative to the control samples

276 (Figure 2). This was due to autofluorescence of GO causing high fluorescent background
277 levels that affected *in vivo* Chl-a quantification (Figure 2). Without algae present, GO
278 autofluorescence at a concentration of 100 mg L⁻¹ was >100 AU in the TG201 algae media,
279 and was even higher >300 AU in MilliQ water (data not shown). In contrast to in MWCNT
280 dispersions, algal cells remained dispersed in the presence of high GO concentrations (Figure
281 3B). Although adsorption of algal cells to GO might occur, this could not be determined with
282 light microscopy due to the small size and stable dispersion of GO in the exposure media.

283

284

285 ***In vitro Chlorophyll-a determination***

286 When using *in vitro* Chl-a determination, no direct fluorescence interference and
287 autofluorescence was observed. However, a loss of fluorescence signal relative to control
288 samples containing only algae was observed for both CNM types with both extraction
289 methods applied (Figure 4). Fluorescence was completely reduced in the presence of 100 mg
290 L⁻¹ MWCNT using the modified acetone extraction (99.5±0.2%) and the EtOH extraction
291 described in the ISO protocol (99.1±0.4%) (Figure 4A). Fluorescence was also decreased in
292 the presence of 100 mg L⁻¹ GO, however, the reduction was significantly less pronounced,
293 being 21.5±4.7% when using the EtOH extraction method and 52±8 % ($p=0.0049$) when
294 using the acetone extraction method (Figure 4B).

295

296 ***Dose dependence***

297 When applying the acetone extraction method, a clear dose dependence of fluorescence
298 reduction was found for both CNMs tested. However, the effects were more pronounced for
299 MWCNT than for GO (Figure 5). A significant reduction in fluorescence of 16.5±4% was

300 observed for the MWCNT at a concentration of only 1 mg L⁻¹ ($P < 0.0001$). At the tested
301 concentrations of 10, 50 and 100 mg L⁻¹ the relative reduction in fluorescence was 65±1.6%,
302 95.5±0.3 and 98±0.1 %, respectively. In the GO exposures, a significant reduction in
303 fluorescence was seen at concentrations of 50 and 100 mgL⁻¹, being 22.4±1.4% and 32±1.1%,
304 respectively.

305

306 *In vitro Chlorophyll-a determination – Extraction point and time dependence*

307 In order to understand more about the adsorption process, the CNMs were added to both high
308 and low algae concentrations at different time points in the extraction process: (i) added to the
309 live algae immediately prior to extraction (TP1), (ii) added directly added after Chl-a
310 extraction with acetone but prior to the 24 h settling period (TP2), and (iii) added after the 24
311 h settling period, 10 minutes before fluorescence measurements were made (TP3). Figure 6
312 summarises the degree of Chl-a adsorption of the two CNMs at both high and low algae
313 concentrations at TP1-TP3. Chl-a fluorescence was significantly decreased (95 - >100 %) in
314 the presence of 100 mg L⁻¹ MWCNT for both low and high algae concentrations compared to
315 controls at all TP. At all TPs, the decrease in fluorescence was more efficient ($P < 0.001$) for
316 the low algae concentration than for the high algae concentration (Figure 6A). Association of
317 Chl-a with the MWCNTs was also observed microscopically (Figure 7). While no
318 fluorescence signal was detected in the MWCNT only controls (Figure 7A and B), the
319 presence of high Chl-a concentrations resulted in a fluorescence signal which was most
320 intense and most stable in association with MWCNT aggregates (Figure 7D). Furthermore,
321 the reduction in fluorescence was lower for the GO than for the MWCNTs at both test
322 concentrations and at all TPs (Figure 6B). However, the results were generally consistent for
323 this material at both concentrations and all TPs, with fluorescence reduction being 50-55% at
324 the low concentration across all TPs and 35-60% at the high concentration. Again, there was a

325 slight trend towards greater fluorescence reduction when moving from TP3 to TP1.

326

327 **Discussion**

328 The current study assessed the applicability of Chl-a determination by fluorescence for algae
329 toxicity assessment of CNMs. Our results show that the assessed *in vivo* and *in vitro* methods
330 were not suitable for use with CNMs. *In vivo* Chl-a fluorescence measurements for CNMs
331 highlighted two key interference processes; (i) shading of algal fluorescence, and (ii)
332 autofluorescence. Shading of the fluorescent signal was the predominant interference process
333 in the case of the MWCNTs. Moreover, MWCNTs were unstable, formed aggregates and
334 settled out of the exposure dispersion removing any co-adsorbed algal cells (Table 1; Figure
335 3A). The co-adsorption of CNTs and other CNMs to algal cells in exposure studies is well
336 documented (Schwab *et al.*, 2011, Long *et al.*, 2012, Sørensen *et al.*, 2015, Mouchet *et al.*,
337 2016, Nolte *et al.*, 2017). Critically, co-adsorption prevents separation of the two matrix
338 elements, e.g. through CNM settling, which may have offered a solution for achieving
339 accurate Chl-a measurements. Some studies have even detected individual CNTs in the
340 cytoplasm of algal cells which may complicate quantification further (Rhiem *et al.*, 2015).
341 Although shading of the algal cell fluorescence is also expected to be an issue in the case of
342 GO, the results show that autofluorescence of this material is the dominant process, having an
343 even greater impact on the ability to measure Chl-a fluorescence accurately. However, there
344 was no clear evidence of co-adsorption between algal cells and GO (Figure 3B). The GO was
345 relatively stable in the exposure dispersions, making it impossible to identify single GO sheets
346 with light microscopy and therefore detect potential co-adsorption of GO and algal cells. The
347 results of the current assessment support those from previous studies which concluded that *in*
348 *vivo* measurement of Chl-a in tests employing MNMs is not feasible due to shading of algal
349 fluorescence, algal cell adsorption processes and autofluorescence in similar wavelengths as

350 Chl-a (in the case of GO).

351

352 Owing to the significant limitations in achieving accurate determination of Chl-a *in vivo*,
353 recent studies have attempted to develop *in vitro* approaches for Chl-a determination
354 (Hartmann *et al.*, 2012, Cerrillo *et al.*, 2016, Hund-Rinke *et al.*, 2016). In the current study,
355 the issues concerning shading of algal fluorescence and autofluorescence during the Chl-a
356 fluorescence measurements were removed when the *in vitro* approaches were applied.
357 However, the concentration-dependent loss of Chl-a fluorescence signal observed in the *in*
358 *vitro* studies indicates that the Chl-a is being removed from the system through adsorption to
359 the CNMs (Figure 5). The results from the studies assessing the two different extraction
360 methods for *in vitro* Chl-a determination from algal cells showed that the presence of 100
361 mgL⁻¹ MWCNT caused a total loss of Chl-a fluorescence signal in the extracted samples
362 relative to controls. Similarly, GO exhibited a reduction in Chl-a fluorescence relative to the
363 control samples, however, the reduction was less than that observed for the MWCNTs.
364 Furthermore, the reduction in fluorescence was less when using the EtOH extraction and
365 filtration procedure compared to the acetone extraction method reported by Hund-Rinke *et*
366 *al.* [8]. This suggests that the partition coefficient for Chl-a between the CNMs and the two
367 organic solvents is different, with the EtOH retaining a greater proportion of Chl-a than the
368 acetone. However, reduction was still around 20% and is therefore sufficiently high that an
369 accurate determination of Chl-a values is not possible. The differences in Chl-a concentration
370 reduction observed between the MWCNT and GO indicate that the presence of significant
371 amounts of oxygen in the GO material reduces the hydrophobicity of the material and
372 therefore its adsorption affinity for algae cells and Chl-a compared to the MWCNTs. This was
373 supported by the increased level of dispersion stability observed for the GO compared to the

374 MWCNT (Table 1).

375

376 A higher adsorption affinity of MWCNTs was also seen in dose-dependence
377 experiments using acetone extraction. Here, the MWCNTs caused a significant reduction of
378 Chl-a signal at a concentration of only 1 mgL⁻¹. A comparable level of reduction was only
379 reached in GO exposure at a concentration of 50 mg L⁻¹. The dose dependent interactions
380 show that the level of Chl-a adsorption will be a function of the algae to CNM ratio, although
381 the current study does not go as far as determining adsorption isotherms similar to those
382 previously reported for specific persistent organic pollutants and CNMs (Chen *et al.*, 2007,
383 Kah *et al.*, 2014, Glomstad *et al.*, 2016, Zindler *et al.*, 2016). Critically, this dose dependent
384 signal reduction of Chl-a means that a CNM concentration-dependent quantity of the
385 extracted Chl-a will be removed from the sample matrix during the removal of the CNMs.
386 This in turn, might lead to a false interpretation of dose dependent CNM toxicity. Despite the
387 lower degree of Chl-a fluorescence signal reduction observed for GO, it is clear that accurate
388 total Chl-a concentrations cannot be determined for either material, highlighting severe
389 limitations with the proposed modifications to OECD 201 for CNMs.

390

391 A more detailed assessment of the adsorption process was conducted in an attempt to
392 identify possible modifications to the *in vitro* method of Hund-Rinke *et al.*, (Hund-Rinke *et*
393 *al.*, 2016) that would reduce or even remove the impact of adsorption processes on the
394 fluorescence measurements. To achieve this, the CNMs were added to both high and low
395 algae concentrations (reflecting algae concentrations at the start and end of the growth
396 period), and at different time points in the extraction process. Irrespective of when the CNMs
397 were added to the test system, significant adsorption was observed for both CNMs at both
398 algae concentrations and at all TPs studied (Figure 6). This shows not only that CNM-algae

399 co-adsorption occurs, but that CNMs readily adsorb the extracted Chl-a. This was confirmed
400 by microscopic observations, in which high Chl-a fluorescence was shown associated with
401 MWCNT aggregates (Figure 7D). The shortest contact time for CNMs and Chl-a used in the
402 current study was 10 min (TP3). However, even within this very short timeframe, Chl-a
403 adsorption at TP3 was observed to be >100% for the MWCNTs at the low algae concentration
404 and approximately 95% for the high algae concentration. The level of Chl-a adsorption was
405 even higher at TP1 and TP2. Although the degree of Chl-a adsorption to GO was significantly
406 lower than that observed for the MWCNTs at all TPs, it was still determined as ~50% for the
407 low algae concentration and ~35% for the high algae concentration at TP3. Again, Chl-a
408 adsorption was typically higher at TP1 and TP2 for GO. The results show that the Chl-a
409 adsorption process occurs extremely rapidly (< 10 min) and therefore further modification of
410 the OECD 201 method proposed by Hunde-Rinke et al., (Hunde-Rinke et al., 2016) does not
411 appear feasible.

412

413 Consistent with most adsorption models (e.g. Langmuir, Freundlich, and the Dubinin-
414 Ashtakhov model based on Polanyi theory) (Yang *et al.*, 2006, Kah *et al.*, 2011, Glomstad *et*
415 *al.*, 2016), the decrease in fluorescence was more efficient ($P < 0.001$) for the low algae
416 concentration than for the high algae concentration at all TPs (Figure 6A) for MWCNT.
417 Although the reduction in fluorescence due to adsorption was lower for the GO than for the
418 MWCNTs at both test concentrations and at all TPs (Figure 6), the level was still greater than
419 30% indicating that reliable Chl-a measurement reflecting accurate algal growth during the
420 test cannot be achieved. Interestingly, the measured reduction in fluorescence for MWCNTs
421 in low algae concentrations was greater than 100% at all TPs (Figure 6A). It is suggested that
422 the MWCNTs also adsorb the NOM present in the exposure media. The adsorption of NOM
423 to CNTs is well known, and the stabilising effect this has on CNT dispersions is the main

424 reason why NOM is increasingly utilised in aquatic ecotoxicity studies with such materials
425 (Cupi *et al.*, 2015, Linard *et al.*, 2015, Cerrillo *et al.*, 2016, Glomstad *et al.*, 2016).

426

427 **Recommendations**

428 Based on the evidence presented in the current study, it is recommended that all MNMs to be
429 subjected to algal growth tests should first undergo a rapid screening test to determine their
430 affinity for Chl-a adsorption. A detailed approach presented in the Supplementary Information
431 is summarised briefly here. A stock dispersion of the test MNM in freshwater algae media
432 should be prepared and added to a solution of extracted Chl-a in algae test media. The
433 resulting sample should be shaken for 10 mins and the MNMs removed by filtration or
434 flocculation and settling. A control sample, containing only Chl-a, should also be prepared
435 and treated in the same way. The fluorescence of the resulting Chl-a should then be measured
436 and the percentage loss of Chl-a to the test MNM determined. It is suggested that Chl-a
437 adsorption >10% renders the method unsuitable for the selected MNM and an alternative
438 method must be used. Where Chl-a adsorption is <10%, the method is considered suitable and
439 can be used as part of the toxicity test.

440

441 Based on the recommendations by Hund-Rinke *et al.* (Hund-Rinke *et al.*, 2016) and
442 the results of the current study, it is suggested that both *in vivo* and *in vitro* fluorescence
443 determination of Chl-a have limitations and do not represent viable standard methods, which
444 need to be applicable to all MNM types. Cell counting using automated systems can also be
445 limited as agglomerates can be in the size range of the algae and systems are not able
446 differentiate between the two under these circumstances. Although the method is very time
447 consuming and therefore costly, it is recommended that manual cell counting remains the only
448 reliable approach currently available for accurately quantifying algal cell growth in acute

449 toxicity methods. Owing to the importance of standardised algae acute toxicity tests from a
450 regulatory perspective, it is also recommended that alternative approaches for the
451 quantification of algal growth is prioritised in the near future.

452

453 **Conclusion**

454 The currently available algae toxicity test methods are not fully suited to the hazard
455 assessment of the MNMs. A number of issues arise during both the exposure period and in the
456 algal growth quantification methods available, including *in vivo* and *in vitro* fluorescence
457 measurements. *In vivo* quantification is limited due to the issue of MNMs shading algal
458 fluorescence and is unable to provide reliable Chl-a concentrations. *In vitro* fluorescence
459 quantification, based on extraction of the Chl-a followed by removal of the test MNMs, has
460 recently been recommended as the best alternative to *in vivo* quantification. Whilst this
461 remains the case for MNMs which do not exhibit a significant level of algae or Chl-a
462 adsorption, the approach is not suitable CNMs and other MNMs which have a strong
463 adsorptive properties. At present, algal growth quantification by cell counting seems to
464 remain the only method which has potential applicability for all MNM types, but this method
465 is very time consuming and not necessarily suitable for standard toxicity tests which need to
466 be time and cost effective.

467

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471 access to light microscopy equipment.

472

473 **Declaration of Interest**

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593

594 **Tables**

595 Table 1. Properties of MWCNT (NM-400) and GO used in this study. The hydrodynamic diameter (HDD) is given as z-average. For HDD,
 596 polydispersity index (PDI) and zeta potential values are presented as mean±SD; *n*=3.

Material	Identifier	Surface area (m ² g ⁻¹)	Size (nm)	HDD (nm)		PDI		Zeta potential (mV)	
				Stock ^e	Exposure ^f	Stock ^e	Exposure ^f	Stock ^e	Exposure ^f
MWCNT	NM-400	254* ^a 189.3±8.1* ^b	11±3/16.2±3.5* ^c 846 ± 446* ^d	3540±727	2480±1450	0.99±0.013	0.92±0.92	0.8±1.6	-20±2.8
GO	GO	n.a.	<100 nm – 1 μm	245±14.6	214±7.22	0.40±0.018	0.37±0.023	-41±0.14	-28±0.25

597 *Data reproduced from Rasmussen et al. (Rasmussen *et al.*, 2014)

598 a: Derived using BET method

599 b: Derived using SAXS

600 c: MWCNT thickness (the two values shown represent independent measurements conducted in two different laboratories)

601 d: MWCNT length

602 e: Stock concentration of CNMs were 2560 mg L⁻¹, in MilliQ (GO) or MilliQ+20 mgL⁻¹ NOM (MWCNT)

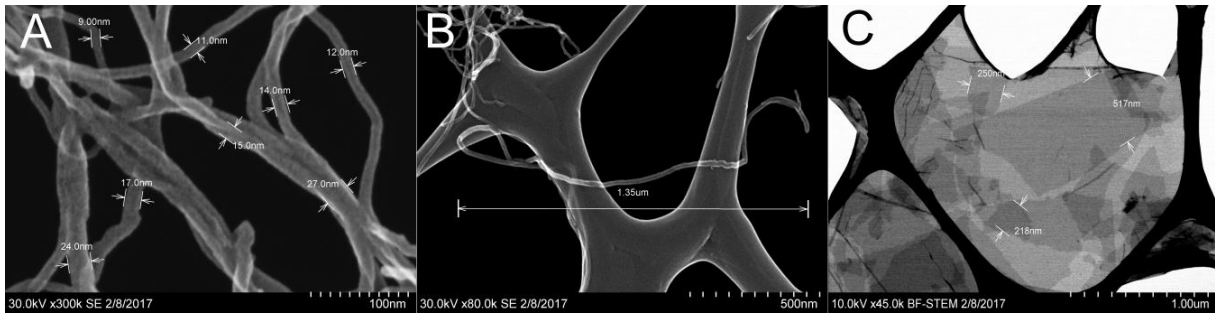
603 f: The exposure concentration of the CNMs was 100 mg L⁻¹ in TG201 (GO) or TG201+10 mgL⁻¹ NOM (MWCNT)

604

605 **Figures**

606

607 Figure 1



608

609

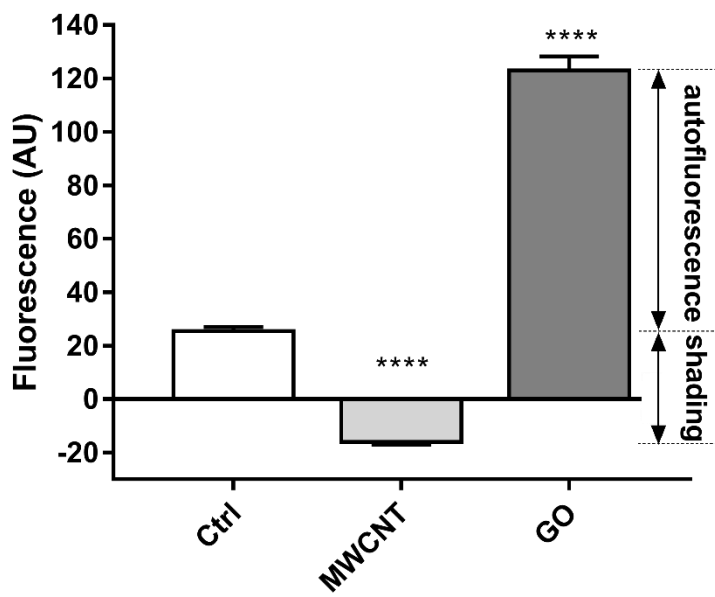
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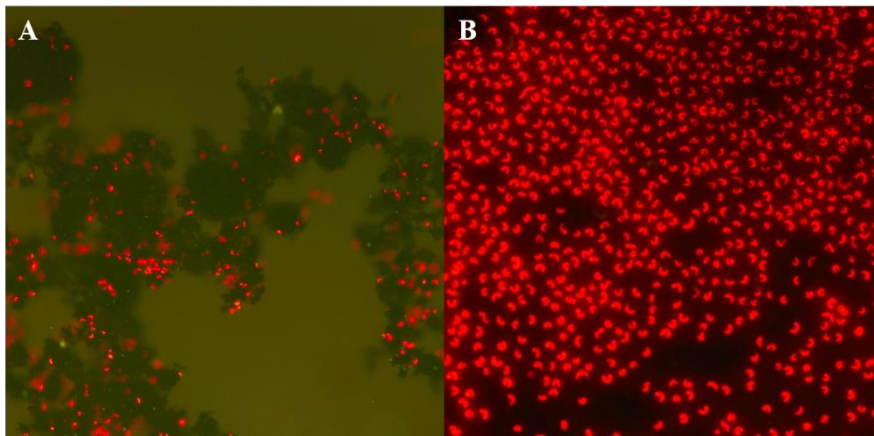
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614 Figure 2



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616 Figure 3



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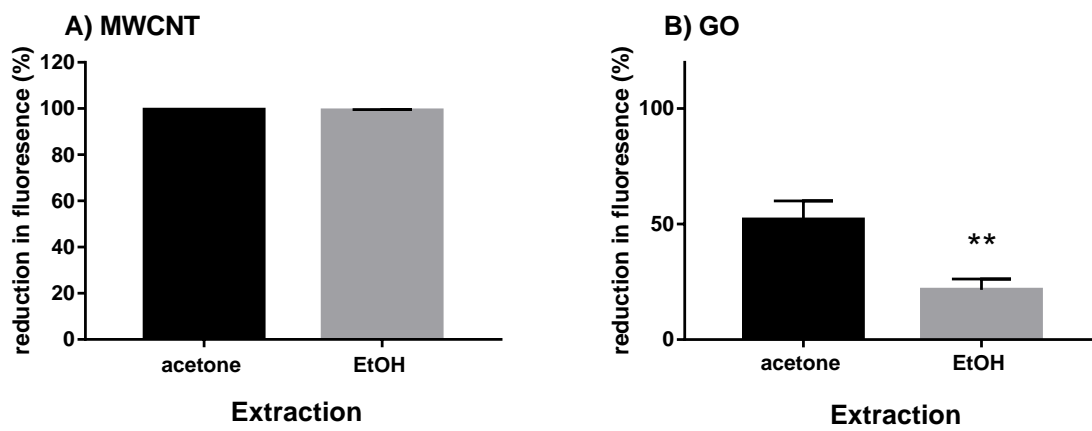
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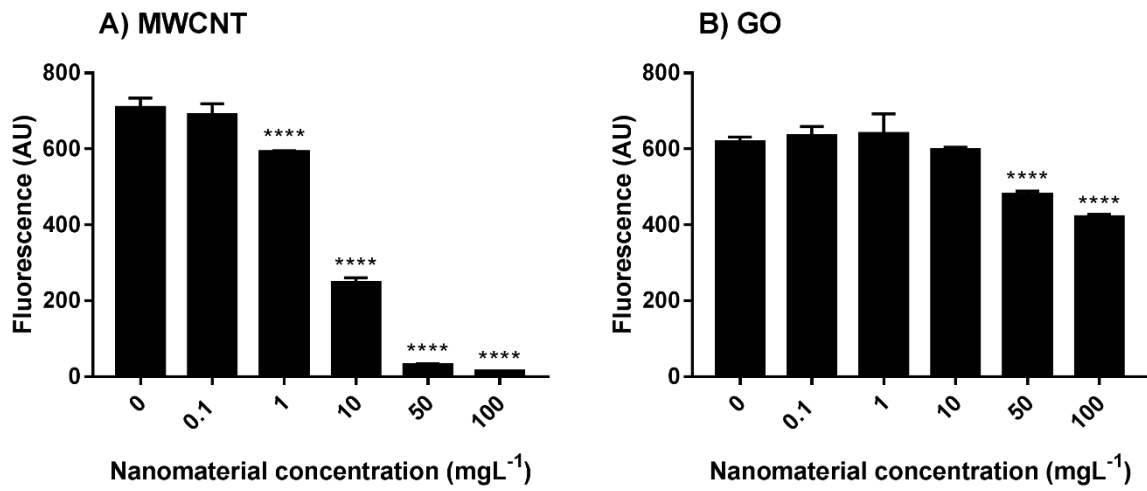
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622 Figure 4



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624 Figure 5



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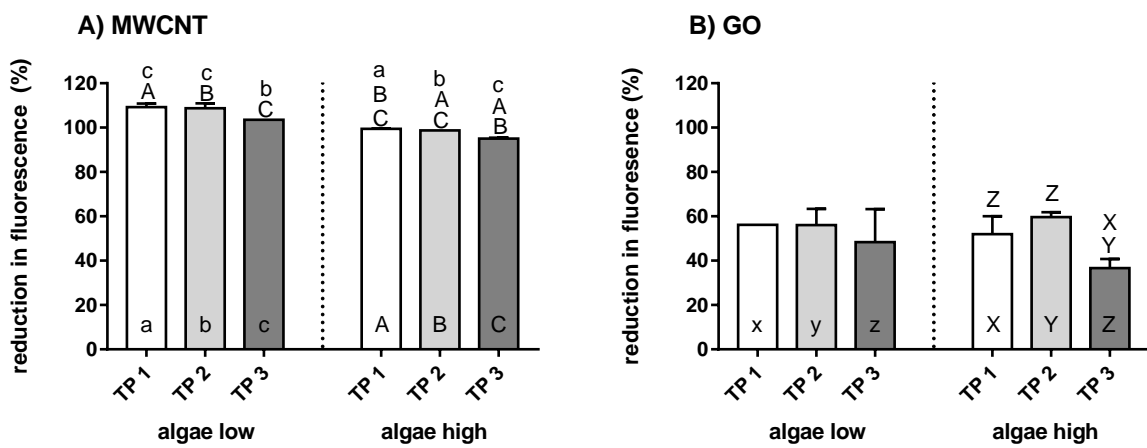
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630 Figure 6

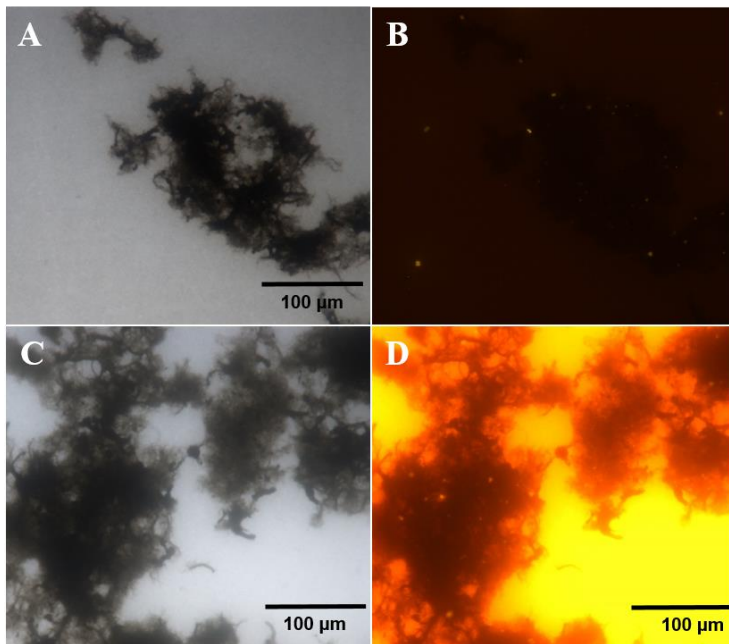


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634 Figure 7



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637

638 *Figure captions*

639

640 Figure 1. Electron microscopic images of A, B) MWCNT in SEM mode and C) GO in STEM
641 mode showing the size of the CNMs used in this study.

642

643 Figure 2. *In vivo* fluorescence measurements of algae in control groups (Ctrl), and in the
644 presence of 100 mg L⁻¹ MWCNT and GO, respectively. The data is shown as mean±SD, *n*=3.
645 Significant differences are indicated (*P*<0.05).

646

647 Figure 3. A) Algae adsorbed to agglomerates of MWCNT (10 mgL⁻¹ in TG201+NOM), and B)
648 Homogenous dispersion of algae in the presence of 50 mgL⁻¹ GO (in TG201).

649

650 Figure 4. Reduction in algal Chl-a fluorescence, relative to control samples containing only
651 algae, after extraction with two different methods: acetone extraction with enhanced
652 flocculation (LBG) and EtOH extraction followed by filtration in the presence of A) 100 mg L⁻¹
653 MWCNT and B) 100 mg L⁻¹ GO added. Data are presented as mean±SD, *n*=3. Significant
654 differences in fluorescence reduction between acetone and EtOH extraction are shown
655 (*P*<0.05).

656

657 Figure 5. Reduction in fluorescence (AU) of Chl-a extracts in the presence of different
658 concentrations of A) MWCNT and B) GO. Chl-a was extracted with acetone and subjected to
659 flocculation and settling in the presence of LBG. Data is shown as mean±SD, *n*=3. Significant
660 differences compared to control groups (0 mgL⁻¹) are indicated (*P*<0.0001).

661

662 Figure 6. Relative reduction of Chl-a fluorescence compared to control using a modified OECD
663 extraction protocol. Fluorescence was determined at two algae concentrations (low and high)
664 in the presence 100 mg L⁻¹ MWCNT (A) and GO (B). Data bars represent samples with addition
665 of CNMs at different time points during the Chl-a extraction: TP1: CNMs added to the live
666 algae; TP2: CNMs added directly added after Chl-a extraction; TP3: CNMs added 10 minutes
667 before fluorescence measurements. The data is shown as mean±SD, *n*=3. Significant
668 differences between groups are indicated by different letters (*P*<0.05).

669

670 Figure 7. Microscope images of MWCNT. A, B) without Chl; C, D) MWCNT mixed with
671 extracted Chl. Fluorescence images (B, D) were taken at the same illumination and microscope
672 settings. Chlorophyll is seen in bright red surrounding the MWCNT. Background Chl is seen
673 as yellow after rapid bleaching during illumination.