



## Single digit parts-per-billion NO<sub>x</sub> detection using MoS<sub>2</sub>/hBN transistors

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### ABSTRACT

2D materials offer excellent possibilities for high performance gas detection due to their high surface-to-volume ratio, high surface activities, tunable electronic properties and dramatic change in resistivity upon molecular adsorption. This paper demonstrates a simple field effect transistor (FET) of molybdenum disulphide (MoS<sub>2</sub>) fabricated on a hexagonal boron nitride (hBN) substrate that can detect NO<sub>x</sub> down to concentrations of 6 ppb and possibly far below at room temperature (RT) with a systematic optimization of the device design and fabrication parameters as well as the device operating conditions. The effects of the substrate, number of MoS<sub>2</sub> layers, channel layout and biasing conditions on the response of MoS<sub>2</sub> FETs to NO<sub>x</sub> were investigated, providing directions for maximizing the sensitivity. This work also sheds light on the issues of recovery and stability and present a methodology for calibration of the sensors which is critical for repeatable and reliable measurements.

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## 1. Introduction

Industrialization and urbanization worldwide has led to increased levels of air-pollutants which can seriously negatively impact human health as well as increase global environment hazards such as ozone layer deterioration, acid rain and photochemical smog [1,2]. These air-pollutants mainly contain harmful gasses such as carbon monoxide (CO), nitrogen oxides (NO<sub>x</sub>) or sulphur dioxide (SO<sub>2</sub>). It has been noted that an exposure to NO<sub>x</sub> concentrations even as low as 53 parts-per-billion (ppb) can potentially damage the human respiration system [3–6]. Existing technology to detect these hazardous gases is based on metal oxides which requires high operating temperatures (> 200 °C) to activate the gas sensing processes (adsorption/desorption), which not only requires fabrication complexity but also leads to high power consumption [7–9].

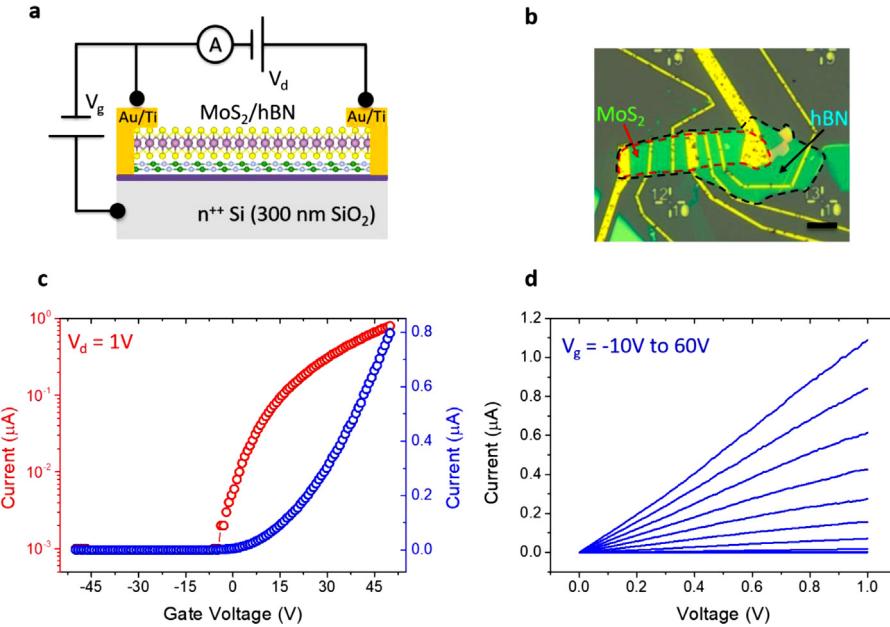
Conversely, two-dimensional (2D) materials have emerged as a very promising class of materials which have been used to develop ultrahigh sensitive and low-power-consumption gas sensors due to their unmatched surface area per unit volume, high surface activities and unique electronic properties [10–14]. Graphene based gas sensors have shown remarkable performance by sensing down to a single gas molecule at room temperature [15]. For high performance gas sensors, transition metal dichalcogenides (TMDs) offer a route to obtain semiconducting properties which is a key parameter, unlike graphene which has a zero bandgap [16–18].

Recent works on MoS<sub>2</sub> based NO<sub>x</sub> sensors have demonstrated the potential of this class of materials with excellent sensitivities ranging from 100 ppm to 8 ppb [19–31]. In these works, elevated temperature, nanoparticles, and red light have been used, which increases the power consumption and complexity and added steps in device fabrication. Given that the United States Environmental Protection agency has stipulated 53 ppb of NO<sub>x</sub> as being dangerous [6], it is imperative to develop low cost highly sensitive NO<sub>x</sub> sensors.

It is known that devices that employ hBN as a substrate are robust and have increased performance due to the reduction of electron-hole puddles and scattering sites given the atomically flat surface [32–34]. In this paper, we employ high quality hBN

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**Fig. 1.** Schematic and optical image (scale bar 10  $\mu\text{m}$ ) of single layer MoS<sub>2</sub>/hBN transistor (a,b). MoS<sub>2</sub>/hBN transistor characteristics (c,d). (c) Transfer characteristics at drain voltage ( $V_d$ ) of 1 V. (d) series of output characteristic curves with different gate voltages ( $V_g$  = -10 V to 60 V).

as a substrate for our mechanically exfoliated MoS<sub>2</sub> devices and demonstrate sensitivity down to at least 6 ppb for NO<sub>x</sub> detection at room temperature through a systematic optimization of the device parameters and operating conditions. Our work provides significant information on the impact of various structural parameters such as type of substrate, number of MoS<sub>2</sub> layers and channel geometry on the response. Moreover, we discuss in detail the sensor biasing conditions and their adjustment to obtain consistent and reliable measurement results as well as maximal sensitivity.

## 2. Experimental methods

### 2.1. Device fabrication

The MoS<sub>2</sub>/hBN Van der Waals heterostructures were fabricated by a dry transfer technique using mechanically exfoliated flakes [35] (details in supporting information Fig. S1). Electron-beam lithography (EBL) was used to pattern the MoS<sub>2</sub> channel with subsequent shaping via reactive ion etching using a mixture of CHF<sub>4</sub> and O<sub>2</sub> gases. Finally, electrodes were realized by EBL and e-beam evaporation and lift-off of Ti/Au (5 nm/100 nm) metals.

### 2.2. Raman characterization

Raman characterization was carried out using a MonoVista CRS+ system with 532 nm laser. The laser power was sufficiently low to avoid sample damage. Additionally, Raman characterization was performed on flake of similar thickness not within the channel of the device.

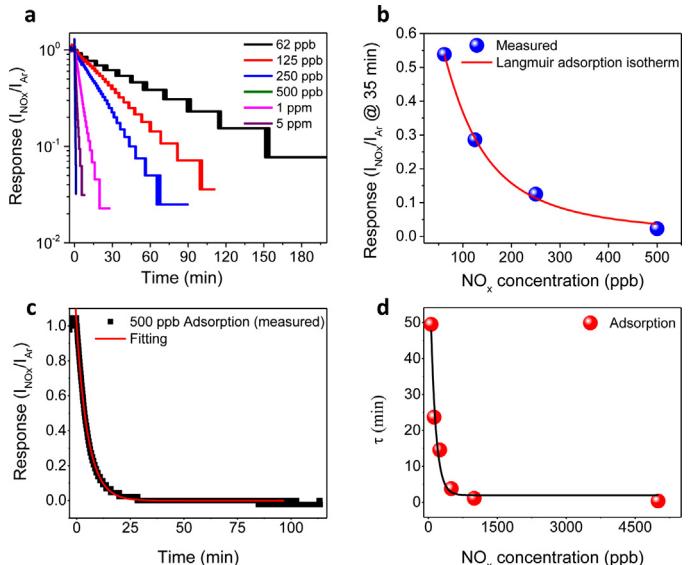
### 2.3. Gas measurements

Gas measurements were carried out in a custom gas chamber at atmospheric pressure. The NO<sub>x</sub> used was a 50/50 mixture of NO<sub>2</sub> and NO. Pre-diluted NO<sub>x</sub> gas bottles were purchased from Linde Gas AS (Norway) with concentration of 50 and 5 ppm balanced by argon. For obtaining desired gas concentration in this work, the gases were further diluted by mixing with argon before supplying

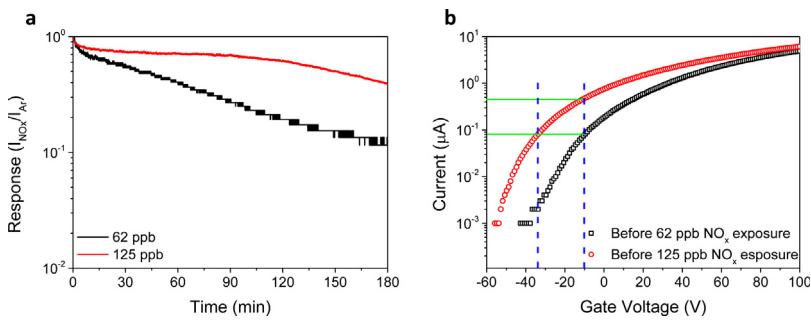
to the gas chamber. The total flow rate of the gases was between 40–80 ml/min.

## 3. Results and discussions

A single layer MoS<sub>2</sub>/hBN field effect transistor (FET) with a channel length of 7  $\mu\text{m}$  and channel width of 10  $\mu\text{m}$  was used as the reference device structure for our work (Fig. 1(a)). The Raman measurement data confirming single-layer thickness of the MoS<sub>2</sub> channel is included in supporting information Fig. S2. An optical image of the device is shown in Fig. 1(b). Fig. 1(c,d) shows the transfer and output characteristics of the fabricated reference device,



**Fig. 2.** Sensor response under the different NO<sub>x</sub> exposures at zero gate voltage ( $V_g = 0 \text{ V}$ ) (a) and NO<sub>x</sub> gas response at fixed time (35 min) for different gas concentrations (62, 125, 250 and 500 ppb) (b). Time dependent response change under the NO<sub>x</sub> exposure of 500 ppb and the corresponding fitted curve using the exponential decay function (c) and extracted adsorption rate constant ( $\tau$ ) from curve fitting for different NO<sub>x</sub> concentrations (d).



**Fig. 3.** Sensor response under the two different  $\text{NO}_x$  exposures (62 ppb and 125 ppb) at -10V gate voltage (a), measured transfer characteristics curves prior to each gas measurements (b).

which was operated in two terminal configuration. As seen in Fig. 1, the device shows n-type behaviour where drain current increases with increasing gate voltage. The field effect mobility extracted from the transfer characteristics is  $1.16 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ .

Fig. 2(a) shows the normalized responses of the device to different concentrations of  $\text{NO}_x$  ranging from 5000 ppb down to 62 ppb at a gate voltage of 0V. The gas response is normalized as  $I_{\text{NO}_x}/I_{\text{Ar}}$ , where  $I_{\text{NO}_x}$  and  $I_{\text{Ar}}$  refer to drain current during the  $\text{NO}_x$  exposure and prior to  $\text{NO}_x$  in Ar atmosphere, respectively. Upon exposure to  $\text{NO}_x$ , the drain current decreases and threshold voltage ( $V_{\text{th}}$ ) shifts toward positive gate voltages, indicating p-type doping of the channel with  $\text{NO}_x$  exposure. These attributes confirm that the gas sensing process is based on charge transfer mechanism, where  $\text{NO}_x$  captures electrons from the  $\text{MoS}_2$  channel. As the concentration of  $\text{NO}_x$  increases, more charges per unit time are transferred from  $\text{MoS}_2$  to  $\text{NO}_x$ , leading to a steeper decrease in the drain current (Fig. 2(a)). Fig. 2(b) shows that our device exhibits good response, capable of detecting very low concentrations of  $\text{NO}_x$  (62 ppb) at room temperature and follows the Langmuir isotherm for molecules adsorbed on the surface (Fig. 2(b)). It therefore confirms that charge transfer is the sensing mechanism [36]. The results also show that the response time is much shorter for higher  $\text{NO}_x$  concentration (20 s for 5 ppm) as shown in Fig. 2(a) and Fig. S4. Furthermore, the adsorption rate constant ( $\tau$ ) of our sensor was extracted which reveals the rate of the  $\text{NO}_x$  adsorption process on the  $\text{MoS}_2$  surface. Fig. 2(c) illustrates the drain current decreases associated with  $\text{NO}_x$  adsorption. The data is fitted with an exponential decay function which confirms that there is only one mechanism associated with  $\text{NO}_x$  adsorption [37]. The adsorption rate constant as the function of different  $\text{NO}_x$  concentrations is presented in Fig. 2(d). It is observed that  $\tau$  decreases as the  $\text{NO}_x$  concentration increases and saturates for the higher  $\text{NO}_x$  concentrations.

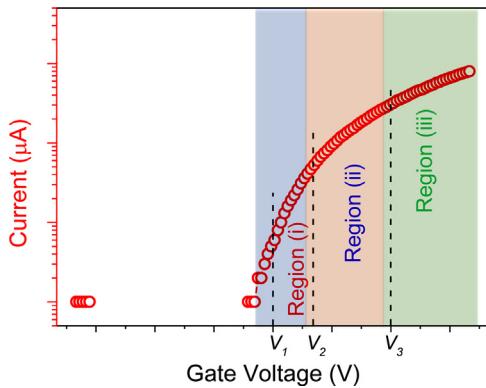
Throughout the rest of the paper, the effects of the device operating conditions and device structure parameters on the response of the device to  $\text{NO}_x$  were systematically investigated. This study will provide an insight into optimizing these parameters to achieve maximal response to  $\text{NO}_x$  and sensor calibration methods in order to obtain consistent and reliable measurement results.

### 3.1. Threshold shift, recovery and gate voltage dependence of the response

As mentioned above,  $\text{NO}_x$  exposure leads to a significant shift of the device's threshold voltage toward positive gate voltages. Various techniques can be employed to recover  $\text{MoS}_2$  gas sensors after  $\text{NO}_x$  exposure. One method is by applying a negative gate voltage pulse to refresh the sensor [38]. The other methods that can be used to recover the sensor after  $\text{NO}_x$  exposure are to anneal at elevated temperatures (100–200 °C) [39–41] or light illumination, where photogenerated holes react with adsorbed gas molecules

and give rise to desorption [42]. The recovery time of the sensor strongly depends on the intensity of the light. Sensor recovery at room temperature by illuminating with white light from a light emitting diode (LED) was demonstrated which shifts the threshold back to original state (Fig. S3). This room temperature sensor recovery suggests that physisorption of  $\text{NO}_x$  molecules on mechanically exfoliated  $\text{MoS}_2$  is the more dominant. In addition, the cyclic test of sensor response and recovery shows good repeatability as shown in supporting information Fig. S4. In addition to gas exposure, environmental effects such as humidity/water molecules could also shift the sensor's threshold voltage [43,44]. A proper calibration of the sensor after each measurement is therefore essential for repeatability and reliability of the measurement results. The following experiment was conducted to study the consequences of uncalibrated operation. We first measured the gate characteristics and then response of the sensor to a  $\text{NO}_x$  concentration of 62 ppb at a gate voltage of -10V. After this measurement, we recovered the sensor using white LED which shifted the threshold voltage back to the original state. Next, the sensor was kept in ambient air for a few days and then the gate characteristics of the sensor was measured again. This was immediately followed by the measurement of the sensor response to a  $\text{NO}_x$  concentration of 125 ppb at the same gate voltage of -10V. As can be seen from the results presented in Fig. 3, the device shows lower response to a concentration of 125 ppb as compared to a concentration of 62 ppb. Lower response to higher concentration is attributed to the threshold shift due to environmental effects before the measurement at a concentration of 125 ppb. This is evident from the transfer characteristics measured right before each gas exposure (Fig. 3(b)). The transfer curve had shifted by -25 V before exposure to 125 ppb of  $\text{NO}_x$ . This shows that by operating the sensor at the same gate voltage of -10 V during the measurement of the response to 125 ppb of  $\text{NO}_x$ , the device is no longer in the same condition that it was when measuring the response to 62 ppb. To restore the same operating conditions as the first measurement with the gate at -10V, the device must be operated with the gate at -35 V during the second measurement.

To study the effect of gate voltage on the sensor response and explore the optimum operating conditions, we measured the response of the reference device to  $\text{NO}_x$  at three different gate voltages. As indicated in Fig. 4, each of the chosen gate voltages corresponds to a different region of the transfer characteristics curve (i.e., regions of different transconductances): i) subthreshold region ( $V_1$ ), ii) transition from subthreshold to linear or quadratic region ( $V_2$ ), and iii) linear region ( $V_3$ ). Due to the threshold shift, the values of  $V_1$ ,  $V_2$ , and  $V_3$  are calibrated before each measurement to ensure that the device is operated at the same condition, i.e., the same doping level. Fig. 5 shows the gate voltage dependence of the response to different concentrations of  $\text{NO}_x$ . The results show that the maximal sensitivity is achieved when the sensor is operated in the subthreshold region ( $V_1$ ). This is because at subthreshold regime, the device is depleted of charge carriers and hence the drain

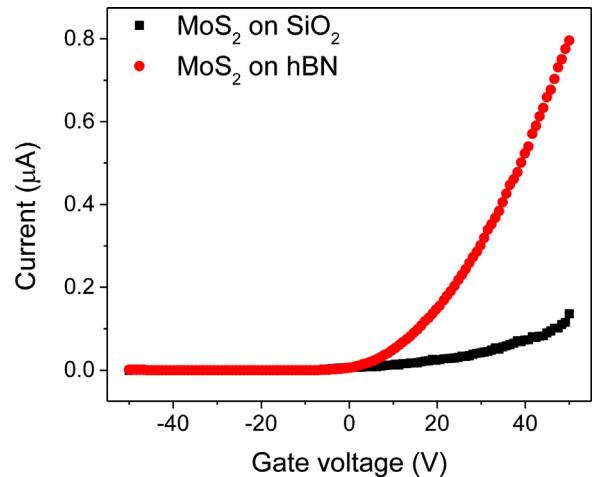


**Fig. 4.** Transfer characteristic curve in logarithmic scale showing different regions (i) subthreshold ( $V_1$ ), (ii) subthreshold to linear ( $V_2$ ), and (iii) linear ( $V_3$ )).

current changes more abruptly upon exposure to a given concentration of NO<sub>x</sub> molecules. However, when the sensor is exposed to the high concentrations of NO<sub>x</sub> (5000 ppb), the effect of the gate voltage becomes negligible. Our experimental results are in agreement with the simulation work presented by Rao *et al.* [17]

### 3.2. Effect of the substrate

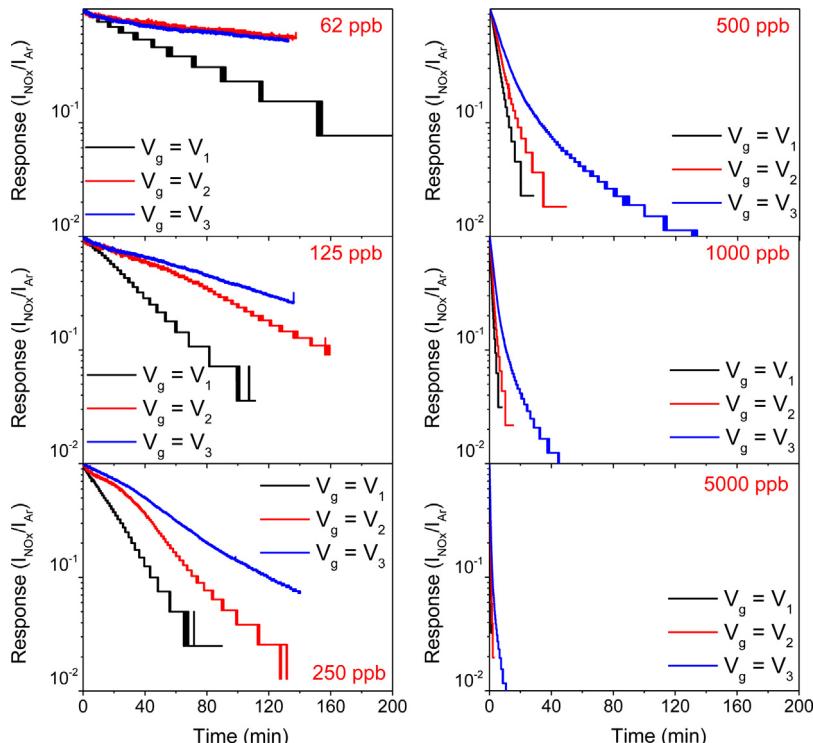
To study the effect of the substrate on the performance of the gas sensor, a single layer MoS<sub>2</sub> transistor on bare SiO<sub>2</sub> (MoS<sub>2</sub>/SiO<sub>2</sub>) was fabricated and compared with the reference device which is on hBN (MoS<sub>2</sub>/hBN). Fig. S5 in the supporting information shows the optical image of the single layer MoS<sub>2</sub> transistor fabricated on SiO<sub>2</sub> substrate. Before conducting sensing comparison, the transistor performance comparison of these devices was carried out. For effective comparison, both devices were chosen with the same dimensions (channel length and width). The transfer characteristic curves of both devices are presented in Fig. 6. The field effect mobil-



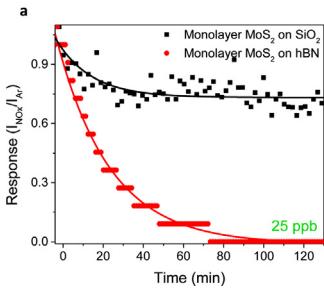
**Fig. 6.** Transfer characteristic curve of single layer MoS<sub>2</sub>/SiO<sub>2</sub> (black curve) and single layer MoS<sub>2</sub>/hBN transistor (red curve), respectively. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).

ity of the devices was extracted as  $1.16 \text{ cm}^2 \text{ V}^{-1} \text{s}^{-1}$  for MoS<sub>2</sub>/hBN and  $0.25 \text{ cm}^2 \text{ V}^{-1} \text{s}^{-1}$  for MoS<sub>2</sub>/SiO<sub>2</sub> transistor, respectively. In addition, the current on/off ratio of  $10^4$  and  $10^3$  was observed for MoS<sub>2</sub>/hBN and MoS<sub>2</sub>/SiO<sub>2</sub> device, respectively. This performance difference can be attributed to fewer interface traps between MoS<sub>2</sub> and hBN compared to MoS<sub>2</sub> and SiO<sub>2</sub>. In addition, hBN is an atomically flat substrate with fewer scattering sites.

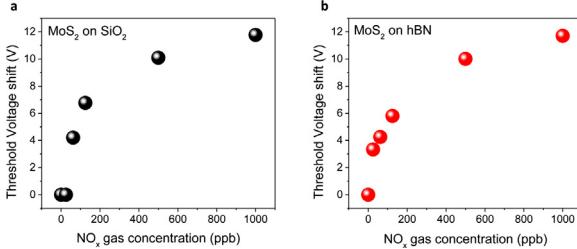
For the comparison of NO<sub>x</sub> gas sensing, both transistors (MoS<sub>2</sub>/hBN and MoS<sub>2</sub>/SiO<sub>2</sub>) were operated in the subthreshold region (Fig. 4), and their responses were measured. Fig. 7 shows that upon exposure to lower concentrations of NO<sub>x</sub> (25 ppb), the MoS<sub>2</sub>/hBN device exhibits higher response as compared to the MoS<sub>2</sub>/SiO<sub>2</sub> device. However, the effect of substrate becomes negli-



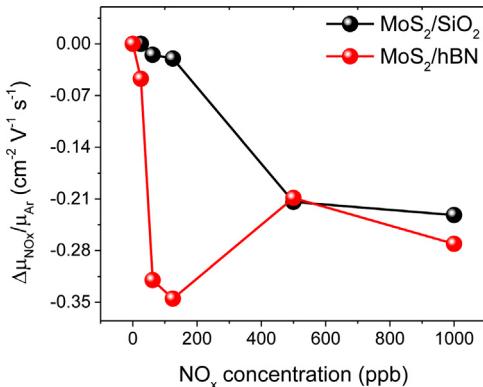
**Fig. 5.** Gate voltage dependent response of the single layer MoS<sub>2</sub>/hBN gas sensor. The sensor was operated at three different gate voltages ( $V_1$ ,  $V_2$  and  $V_3$ ) and the response to various NO<sub>x</sub> exposures (62, 125, 250, 500, 1000 and 5000 ppb) was measured.



**Fig. 7.** Effect of the substrate on sensor response. Gas sensing response of MoS<sub>2</sub>/SiO<sub>2</sub> and MoS<sub>2</sub>/hBN devices under various NO<sub>x</sub> exposures (25 and 500 ppb). The sensors were operated at subthreshold gate voltages.



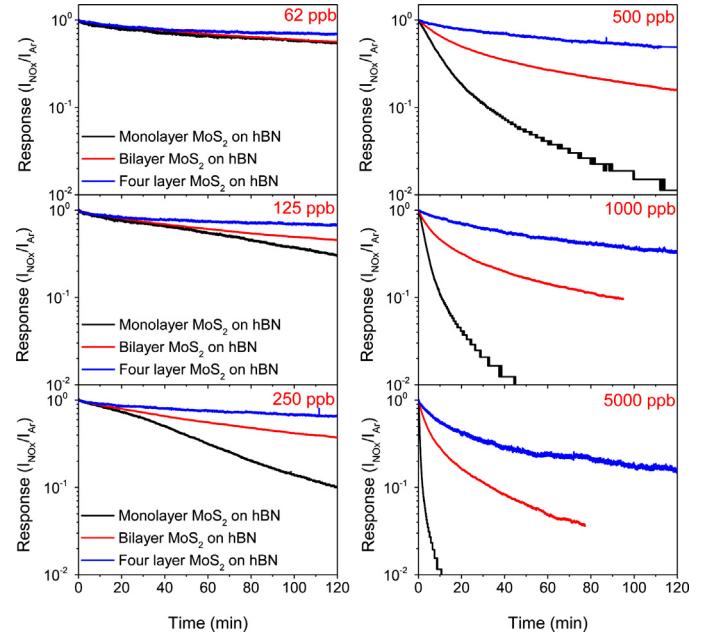
**Fig. 8.** Threshold voltage shift upon different gas exposures for MoS<sub>2</sub>/SiO<sub>2</sub> device (a) and for MoS<sub>2</sub>/hBN device (b), respectively.



**Fig. 9.** Effective mobility change as a function of NO<sub>x</sub> concentrations (25, 62, 125, 500 and 1000 ppb) for MoS<sub>2</sub>/SiO<sub>2</sub> device (black curve) and MoS<sub>2</sub>/hBN device (red curve). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).

gible when the device is exposed to the higher NO<sub>x</sub> concentrations (500 ppb).

To understand the mechanism behind the substrate effect, transfer characteristic curves were recorded after each gas measurement and the corresponding threshold voltage shift and change in the mobility for the MoS<sub>2</sub>/SiO<sub>2</sub> device and MoS<sub>2</sub>/hBN device were extracted and shown in Figs. 8 & 9. A relatively large threshold voltage shift is observed for the MoS<sub>2</sub>/hBN device at low NO<sub>x</sub> concentrations (25 ppb) while for higher NO<sub>x</sub> concentration, almost same threshold voltage shift is observed for both devices. The threshold voltage shift reflects the amount of charge transfer upon gas exposure; hence, it suggests that the MoS<sub>2</sub>/hBN device is more sensitive to ultra-low concentrations of NO<sub>x</sub> (25 ppb). Furthermore, a change in mobility is also observed for both devices under NO<sub>x</sub> exposure (Fig. 9). In the case of MoS<sub>2</sub>/hBN, a sharp decrease in the mobility is observed for low concentration gas exposures. This can be explained by the increase in scattering sites on adsorption of NO<sub>x</sub> molecules due to device being more homogeneous on a flatter substrate. However, in case of the MoS<sub>2</sub>/SiO<sub>2</sub> device, the change



**Fig. 10.** Effect of the MoS<sub>2</sub> layer thickness on sensor response. Gas sensing response of monolayer, bilayer and four-layer MoS<sub>2</sub> devices under various NO<sub>x</sub> exposures (62, 125, 250, 500, 1000 and 5000 ppb). The sensors were operated at saturated regime ( $V_g = 30$  V) in order to keep the doping level same.

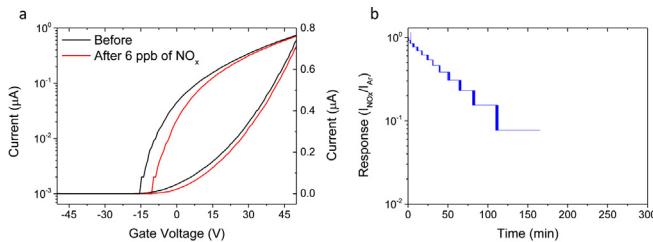
in mobility in the presence of NO<sub>x</sub> is much lower for low concentrations which shows that effect of surface roughness is more dominant. This can be seen from the pristine device characteristics (Fig. 6), where the MoS<sub>2</sub>/hBN device exhibits high mobility and current on/off ratio as compare to MoS<sub>2</sub>/SiO<sub>2</sub>.

### 3.3. Effect of number of layers

The effect of MoS<sub>2</sub> layer thickness on the response of the sensor to NO<sub>x</sub> was also studied. In addition to the reference device, we also fabricated FETs with bilayer and four-layer MoS<sub>2</sub> on hBN. The transfer characteristics curves of the devices with different number of MoS<sub>2</sub> layers are shown in Fig. S6. The threshold voltage is different for FETs with different number of MoS<sub>2</sub> layers as the carrier concentration varies with MoS<sub>2</sub> thickness. Hence, operating the devices at same gate voltage would not provide a correct comparison of their response to NO<sub>x</sub>. Instead, the devices were operated at different gate voltages that lead to the same doping level or the same effective point of operation on the transfer characteristics curve as discussed in Section 3.1. we operated all three sensors at the linear regime of the transfer characteristics curve ( $V_g = 30$  V) and measured the response of the devices under various NO<sub>x</sub> gas exposures as shown in Fig. 10. Our results show that the response decreases with increasing number of MoS<sub>2</sub> layers. This can be attributed to the screening effect [45]. In monolayer MoS<sub>2</sub> FET, NO<sub>x</sub> interacts with the surface charge carriers as the channel is monolayer. In the case of multi-layer MoS<sub>2</sub>, on the other hand, NO<sub>x</sub> interacts less with carriers further away from the top layer. Furthermore, as can be seen in Fig. 10, the effect of number of MoS<sub>2</sub> layers becomes more significant at higher gas concentrations where the response differs by orders of magnitude.

### 3.4. Effect of channel dimensions

To demonstrate the effect of MoS<sub>2</sub> channel dimensions on response, we fabricated hall bar shaped channel (Fig. S7). For comparison, we used two different channel lengths (CHL1 = 12  $\mu\text{m}$  and CHL2 = 24  $\mu\text{m}$ ) with same channel width (4  $\mu\text{m}$ ) and measured the



**Fig. 11.** Sensing of a single digit ppb of  $\text{NO}_x$ . Transfer characteristic curve of single layer  $\text{MoS}_2/\text{hBN}$  device before and after of an ultra-low exposure of  $\text{NO}_x$  (6 ppb) (a). Time response of single layer  $\text{MoS}_2/\text{hBN}$  device under an ultra-low exposure of  $\text{NO}_x$  (6 ppb) (b). The sensor was operated at sub-threshold gate voltage of  $-5\text{ V}$ .

electrical characteristics for each channel length in order to choose the same operating conditions for the devices. The transfer curves of both channel lengths show the same threshold voltage (Fig. S8), and therefore operating the sensors at same gate voltage gives a correct comparison. Fig. S9 shows the response of the devices with CHL1 and CHL2 under fixed gate bias of  $-10\text{ V}$  to various  $\text{NO}_x$  gas concentrations (62, 125, 250, 500, 1000 and 5000 ppb). The device with shorter channel length (CHL1) shows a relatively higher response as compared to long channel (CHL2). This is expected as the charges undergo less scattering during transport through a shorter channel.

### 3.5. Maximal sensitivity

The results obtained from the above studies allowed us to choose the best parameters for maximum sensitivity. We have demonstrated that higher sensitivities to  $\text{NO}_x$  can be achieved with monolayer  $\text{MoS}_2$  on hBN substrate with short channel length and by operating at the subthreshold regime. By implementing all optimizations based on studies reported in the previous sections, single digit ppb sensitivity to  $\text{NO}_x$  was obtained. Fig. 11(a) shows the transfer characteristics before and after the exposure of the optimized sensor to 6 ppb of  $\text{NO}_x$ . A significant shift in the transfer curve was observed even after such low exposures. Moreover, one order of magnitude change in the resistance of  $\text{MoS}_2$  was observed within two hours of ultra-low  $\text{NO}_x$  exposure (6 ppb) as shown Fig. 11(b). The signal to noise ratio is one the important parameters for the high-performance gas sensor, especially for low concentration gas sensors. Our sensor shows an order of magnitude change in resistance at ultra-low exposure (6 ppb). Our experimental setup is currently not capable of providing gas concentrations below 6 ppb, and hence we were not able to explore the ultimate  $\text{NO}_x$  detection limit of  $\text{MoS}_2$  transistors. However, the remarkable response to 6 ppb of  $\text{NO}_x$  indicates that the detection limit can be well-below 6 ppb.

## 4. Conclusion and outlook

We performed a comprehensive study on the response of  $\text{MoS}_2$  transistors to  $\text{NO}_x$  exposure. In this study, we demonstrated how the device structure parameters and operating conditions impact the response. Our findings show that the response of the  $\text{MoS}_2$  transistors to  $\text{NO}_x$  can be significantly enhanced by using hBN substrate instead of  $\text{SiO}_2$ , using monolayer  $\text{MoS}_2$  channel, and reducing the channel length. The bias voltage used for the sensor operation also significantly affects the response and our results show that the maximal response is obtained when the device is operated in the subthreshold regime of the transfer characteristics curve. A known characteristic of  $\text{MoS}_2$  transistors as gas sensor is their poor self-recovery at room temperature, which was also observed in our work. We have shown that it is extremely important to calibrate the device by adjusting the bias voltage before each operation in

order to assure that the measurement results are consistent and reliable. We have demonstrated that  $\text{MoS}_2/\text{hBN}$  transistors with a careful optimization can detect  $\text{NO}_x$  concentrations as low as 6 ppb (or possibly even significantly lower, which we are not capable of testing due to our experimental setup constraints).

This work was focused on demonstrating ultra-high sensitivity of bare  $\text{MoS}_2/\text{hBN}$  transistor to  $\text{NO}_x$  at room temperature. We also addressed the issue of recovery. There are also other key parameters that need to be considered for the development of high-performance  $\text{NO}_x$  sensor such as selectivity, which is beyond the scope of this paper. One method to enhance selectivity to certain gas species is to decorate the  $\text{MoS}_2$  transistors with nanoparticles of specific work functions which is currently under investigation by our group.

## CRediT authorship contribution statement

**Ayaz Ali:** Conceptualization, Methodology, Investigation, Writing - original draft, Writing - review & editing, Visualization. **Ozhan Koybasi:** Conceptualization, Methodology, Validation, Writing - review & editing, Supervision. **Wen Xing:** Methodology, Investigation, Resources. **Daniel N. Wright:** Methodology, Investigation, Resources. **Deepak Varandani:** Project administration, Funding acquisition. **Takashi Taniguchi:** Methodology, Resources, Writing - original draft. **Kenji Watanabe:** Methodology, Resources, Writing - original draft. **Bodh R. Mehta:** Conceptualization, Methodology, Writing - review & editing, Supervision, Funding acquisition. **Brandon D. Belle:** Conceptualization, Methodology, Validation, Writing - review & editing, Supervision, Funding acquisition.

## Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:<https://doi.org/10.1016/j.sna.2020.112247>.

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